

# Mount Kemble Lake

2014 Year End Water Quality Summary Mount Kemble Lake Association, Inc. Morristown, NJ January 9, 2014

Ruth Chaney Mount Kemble Lake Association 28 Lake Trail West Morristown, NJ 07960

#### YEAR END SUMMARY 2014 WATER QUALITY PROGRAM MOUNT KEMBLE LAKE

#### INTRODUCTION

The following report is the 2014 Year-End Summary of the Water Quality Monitoring and Lake Management Program for Mount Kemble Lake located in Morristown, Morris County, New Jersey. This report includes details of lake surveys, the water quality monitoring program, phytoplankton surveys, and observations logged during site visits in 2014. Recommendations for Mount Kemble Lake management efforts for the 2015 season are also included. The Appendix of this report includes graphs and tables of the 2014 field data, reference guides, along with supporting documents for this report.

The 2014 Lake Management Program for Mount Kemble Lake focused on control of nuisance and invasive aquatic plant growth, most specifically curly-leaf pondweed (*Potamogeton crispus*), leafy pondweed (*Potamogeton foliosus*), and southern naiad (*Najas guadalupensis*).

Scientific Name	Common Name
Potamogeton foliosus	Leafy Pondweed
Potamogeton crispus	Curly-leaf Pondweed
Lemna minor	Small Duckweed
Najas guadalupensis	Southern Naiad
Potamogeton diversifolius	Variable Pondweed

Table 1. 2014 Observed Aquatic Macrophytes.

Through the season a total of five (5) different aquatic macrophytes were observed during surveys of the lake (Table 1), with invasive species highlighted in red. One (1) of these species duckweed (*Lemna minor*) is a floating aquatic plant.

#### WEATHER DISCUSSION

In 2014, the mid-Atlantic region saw below to well below average temperatures (Rutgers Climate Lab Figure 1.) during January through April, following what many recall as being the coldest winter since the 1970's. Numerous snowfall records were also shattered in many areas. The cold temperatures that extended well into April influenced generally slow development of aquatic vegetation growth through the spring season, with aquatic plant growth not developing until May. The remainder of the management season experienced mostly average atmospheric temperatures, with August becoming approximately two degree Fahrenheit below average for the month. Overall, precipitation amounts through the season were relatively average, until dry

NJ Monthly Temperature Departures (October 2013 - September 2014) artures calculated from differences between observed monthly temperatures and 1981-2010 monthly average: 6.0 Records based on observations dating back to 1895 5.0 4.0 3.0 2.0 Departure (°F) 1.0 0.0 -1.0 -2.0 -3.0 -4.0 -5.0 -6.0 g Nov ដ្ឋ JAN 盟 MAR A PR MAY S Ę AUG ß Figure 1: Temperature Departures. Month NJ Monthly Precipitation Departures (October 2013 - September 2014) en observed monthly precipitation and 1981-2010 monthly averages 3 Records based on observations dating back to 1895 2 1 Departure (inches) 0 -1 -2

conditions persisted for much of August and September in northern New Jersey (Rutgers Climate Lab Figure 2.).

## LAKE MANAGEMENT

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Figure 2: Precipitation Departures.

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Aquatic biologists were at Mount Kemble Lake on seven (7) dates from April through August to conduct on-water assessments of aquatic vegetation and algae growth, and to perform *in situ* water quality analysis. On three (3) dates, comprehensive water quality analysis was conducted including, sampling for planktonic algae and zooplankton, lab sample collection and lake profile analysis for temperature and dissolved oxygen. Following each survey, biologists would review lake conditions to determine if management activity was required. In 2014, on two (2) dates

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Allied Biological field staff conducted herbicide and algaecide applications for control of nuisance and invasive aquatic vegetation growth. The table below provides a reference to indicate dates of applications, what aquatic pesticides were applied, and the target acreage and aquatic plant species for each date (Table 2). Additionally, on July 23<sup>rd</sup> an application of Aluminum Sulfate was conducted to neutralize nutrients within the water column.

Date	Service Performed	Acres Treated	Target Species
6/6/2014	Copper Sulfate	5	Filamentous algae
6/6/2014 Reward		5	P. foliosus / P. crispus
7/23/2014	Aluminum Sulfate	13.6	Total Phosphorous
8/20/2014	Copper Sulfate	5	Filamentous algae
0/20/2014	Reward	5	N. guadalupensis

Table 2: Mount Kemble Lake 2014 Treatment Log

Early season surveys conducted at Mount Kemble Lake during April and May were highlighted by only a single rooted stem of curly-leaf pondweed (*P. crispus*) along the eastern shoreline, and no other submersed aquatic plant growth observed. Filamentous algae densities were also light in April with only scattered trace surface "clumps" along the lake perimeter, while there was no filamentous algae growth observed during the May  $23^{rd}$  lake survey. Through this time shoreline surveys of Clubhouse Pond indicated that generally sparse density filamentous algae growth was scattered along the lake bottom and surface. An application of the contact herbicide Reward (a.i. diquat) was performed on June 6<sup>th</sup> targeting a five acre community of leafy pondweed (*P. foliosus*) and curly-leaf pondweed that were growing collectively at sparse density along most of the western shoreline, and across the shallow inlet area of Mount Kemble Lake. In addition to the herbicide management, an application of the algaecide Copper Sulfate was also conducted within the same treatment area for usually sparse density surface filamentous algae growth. Water clarity at this time also increased to eight feet, from 4 feet and 3.5 feet observed in April and May, respectively.

A lake survey performed on July  $3^{rd}$  indicated that scattered patches of duckweed (*L. minor*) were located along the shoreline, along with a moderate density area of variable pondweed (*P. diversifolius*) north of the boat launch, and sparse patches along the eastern and western shorelines. Usually light density filamentous algae growth was colonizing on the plants that were decomposing from the previous herbicide treatment. An application of Aluminum Sulfate was performed on July  $23^{rd}$  to target nutrients in the water column by binding the material along with other organic and inorganic matter and settling the "floc" to the bottom of the lake. The Aluminum Sulfate neutralizes nutrients rendering them unavailable, limiting plant and algae development. A light density bloom of unicellular algae was visible in the water column at this time, reducing water clarity to a little more than five feet. Growth of southern naiad was also established at sparse to moderate density along most of the inlet area and western shoreline.

The final lake management activity of the season was conducted on August 20<sup>th</sup> with an application of Reward and Copper Sulfate performed to control nuisance growth of southern naiad and filamentous algae across the inlet and adjacent shoreline areas. Water clarity improved to eight feet at this time as planktonic algae growth was only observed at trace density in the water column.

#### WATER QUALITY MONITORING PROGRAM

In 2014, the water quality monitoring program included *in-situ* field measured limnological analysis including temperature/dissolved oxygen profiles, pH, transparency, alkalinity, and total hardness. In addition, surface water chemistry samples were collected at the north inlet and lake station, as well as from the lake bottom at the lake station site, and transported to Aqua Pro-Tech Laboratories (Fairfield, New Jersey) for analysis of the following parameters: ammonia, conductivity, nitrate, total phosphorus, and total suspended solids. On April 28<sup>th</sup>, samples for total phosphorous were also collected at the inlet and outlet in accordance with the New Jersey Total Maximum Daily Load (TMDL) threshold. Collection for phytoplankton and zooplankton identification and enumeration was also performed on three dates. Provided in the Appendix is a short description of each water quality parameter, and laboratory data results. Below is the water quality data tabulated to provide a seasonal reference.

Mount Kemble Lake Water Quality Results – Inlet Station					
Parameter Units 4/28/2014 Limits					
Total Phosphorus mg/L 0.01 0.03					
	Units	Units 4/28/2014			

WATER QUALITY DATA TABLES

Table 3. 2014 Mount Kemble Lake Water Quality Results

Mount Kemble Lake Wat	Mount Kemble Lake Water Quality Results- North Station				
Parameter	Units	4/28/2014	6/6/2014	7/302014	Limits
Temperature	С°	14.7	21.2	26.4	NA
Dissolved Oxygen	mg/L	13.13	8.12	4.9	<4.0
ph	SU	8.0	7.5	7.5	9
Alkalinity	mg/L	44	76	60	NA
Total Hardness	mg/L	120	80	120	NA
Secchi	feet	4′	8′	5′	<4′
Ammonia	mg/l	ND	ND	ND	0.3
Nitrate	mg/L	0.9	0.98	ND	0.3
Total Phosphorus	mg/L	0.03	0.05	0.01	0.03
Total Suspended Solids	mg/L	ND	ND	ND	25
Conductivity	Umhos/cm	320	300	291	1500

Table 4. 2014 Mount Kemble Lake Water Quality Results

Results highlighted in red identify those that exceed the acceptable lake management limit.

Mount Kemble Lake Wat	Mount Kemble Lake Water Quality Results- Lake Station Surface				
Parameter	Units	4/28/2014	6/6/2014	7/302014	Limits
Temperature	°C	14.8	21.6	24.9	NA
Dissolved Oxygen	mg/L	13.63	9.51	8.55	<4.0
ph	SU	8.0	8.0	7.5	9
Alkalinity	feet	44	84	60	NA
Total Hardness	mg/L	120	100	100	NA
Secchi	mg/L	4′	8.5′	5′	<4'
Ammonia	mg/l	ND	ND	ND	0.3
Nitrate	mg/L	0.9	0.98	ND	0.3
Total Phosphorus	mg/L	0.03	0.03	0.02	0.03
Total Suspended Solids	mg/L	ND	ND	ND	25
Conductivity	Umhos/cm	310	290	291	1500

Table 5. 2014 Mount Kemble Lake Water Quality Results

Results highlighted in red identify those that exceed the acceptable lake management limit.

Mt. Kemble Lake Water 0					
Parameter	Units	4/28/2014	6/6/2014	7/302014	Limits
Dissolved Oxygen	mg/L	3.34	.08	0.38	<4.0
Ammonia	mg/L	ND	0.48	ND	0.3
Nitrate	mg/L	0.95	ND	ND	0.3
Total Phosphorus	mg/L	0.03	0.08	0.02	0.03
Total Suspended Solids	mg/L	ND	6.00	ND	25
Conductivity	umhos/cm	317	325	289	1500

Table 6. 2014 Mount Kemble Lake Water Quality Results

Results highlighted in red identify those that exceed the acceptable lake management limit.

Mount Kemble Lake Water Quality Results – Outlet Station					
Parameter Units 4/28/2014 Limits					
Total Phosphorus mg/L 0.03 0.03					

Table 7. 2014 Mount Kemble Lake Water Quality Results

#### WATER QUALITY RESULTS SUMMARY

During 2014, the surface water temperature was 14.8° C in April, and increased to 21.6°C, and 24.9°C in June and July, respectively. The water temperature on the July 3<sup>rd</sup> lake survey date was the highest observed temperature of the season at 28.0° C. The pH values collected from the inlet and lake station sites throughout the year ranged from 7.5 to 8.0, which falls within the typical range for freshwater lake systems, and is within historical readings of the past several years for Mt. Kemble Lake. The hardness levels ranged from 80 mg/L to 120 mg/L at either location throughout the season. The typical range characteristics to freshwater lakes are those falling between 4 and 200mg/L. The chemical composition of Mount Kemble Lake's surface water is considered moderately hard water. The alkalinity values remained consistent throughout the year from 44 to 84 mg/l, and within a comparable level compared to similar NJ freshwater lakes' chemical composition. Conductivity was consistent throughout the season with values

ranging from 289 to 320  $\mu$ mhos/cm., with the highest observed value obtained in the June lake bottom sample.

Ammonia and nitrates are nutrients based on the chemical composition of nitrogen. These naturally occurring compounds when influenced by human activity can cause excessive plant and algae growth. The only sample that returned a value for ammonia was the lake bottom sample on the June 6<sup>th</sup> sampling date. The ammonia level at this depth and on this date was a result of the water column being nearly depleted of dissolved oxygen below approximately twelve feet. When dissolved oxygen is exceptionally low, ammonia cannot be oxidized into nitrate, which is why this sample was also the only to return a non detected value for nitrate.

Nitrates were found elevated in the April and June surface samples, but were not detected in the July surface samples. In April, nitrate levels were 0.9 mg/L in both surface sites, and were 0.95 mg/L in the bottom sample. During the June sampling nitrate levels were 0.98 at the lake station and inlet stations. These values are three times the threshold value for excessive aquatic plant productivity. The lack of nitrate in the July samples is a result of the application of Aluminum Sulfate being conducted only seven days prior to sampling depleting the water column of available nutrients.

Total phosphorus is usually present in freshwater lakes at low concentrations. Total phosphorus concentrations in a freshwater lake system should be less than 0.03 mg/L to prevent higher productivity. The only surface sample that returned a value higher than the 0.03mg/L threshold was the June sample at the North Station, which was 0.05 mg/L. In June, the lake station bottom sample was elevated at 0.08 mg/L, while the other two dates were at or below the 0.03 mg/L threshold. In terms of total phosphorous levels, the values observed in 2014 were significantly lower than those observed in 2013, when phosphorous sampling returned values of 0.04 mg/L, 0.09 mg/L and 0.13 mg/L. This translates to an average phosphorous value in 2014 that was half than 2013, declining from 0.086 mg/L to 0.043 mg/L. This appears attributable to the application of Aluminum Sulfate in July that prevented elevated phosphorous levels through the latter half of the summer season.

Oligotrophic	Mesotrophic	Eutrophic	Hypereutrophic
<0.012mg/L	0.012 - 0.024mg/L	0.025 - 0.096mg/L	>0.096mg/L
Very Good	Good	Fair	Impoundments

Table 8: Trophic Status Based on Phosphorus Values

Transparency (water clarity) displayed moderate variability in 2014, with observed secchi readings between 3.5 and 8.5 feet. The highest secchi measurement was recorded on June  $6^{th}$ , while the lowest observed clarity reading was on May  $23^{rd}$ , likely affected by turbidity from recent rainfall. Overall, water clarity was considered fair to good throughout the season, and was not observed to be negatively affected by blooms of planktonic algae.

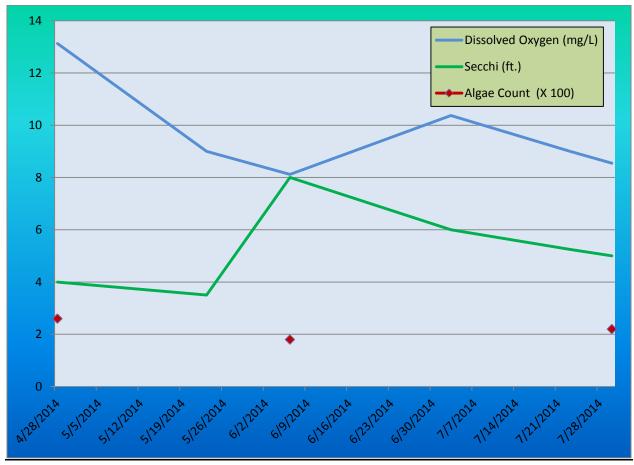


Figure 3. Mount Kemble Lake Seasonal Profile

#### LAKE PROFILE DESCRIPTION

In 2014, temperature and dissolved oxygen profiles were collected from the lake station sampling site utilizing a temperature / dissolved oxygen meter. The April profile revealed a well mixed water column, with favorable dissolved oxygen to a depth of twenty-four feet. During June, the lake profile revealed what is called a positive heterograde curve, which simply means that the water quality conditions of the lake depleted dissolved oxygen below a

	4/28/2014		6/6/2014		7/30	/2014
Depth	Temp.	DO	Temp.	DO	Temp.	DO
(m)	(°C)	(mg/L)	(°C)	(mg/L)	(°C)	(mg/L)
Surface	14.8	13.63	21.6	9.51	24.9	8.55
1	14.4	13.85	21.6	9.35	24.7	8.41
2	14.2	13.56	21.6	9.41	24.6	8.45
3	13.0	14.69	16.9	14.06	24.6	8.38
4	9.3	8.87	12.8	2.90	24.5	8.30
5	7.7	5.31	9.7	0.19	23.8	4.69
6	7.0	5.04	8.1	0.13	21.9	1.65
7	6.5	4.26	7.2	0.11	18.7	0.38
8	6.2	3.34	6.6	0.10		
9			6.6	0.10		

Table 9. Mount Kemble Lake Dissolved Oxygen Profiles.

depth of approximately twelve feet, and likely supported an algae bloom at about nine feet,

which corresponds with the observed water clarity of a little less than nine feet. The algae bloom at this depth is indicated by the spike in dissolved oxygen at this depth alone. This type of water quality condition is observed most frequently in lakes where the surface area is small relative to the maximum depth and protected from intense wind action by surrounding topography and vegetation, which is descriptive of Mt. Kemble Lake. The July 30<sup>th</sup> lake profile displayed a more typical mid-summer stratification, where dissolved oxygen is consistent to a particular depth, in this case approximately twelve feet, and then declines to anoxia at an approximate depth of eighteen feet. Complete profile graphs are provided in the Appendix of this report.

#### **PLANKTON SURVEYS**

Phytoplankton and Zooplankton surveys were conducted at Mount Kemble Lake in conjunction with the water quality monitoring program. In 2014, surface phytoplankton samples were collected at two established water quality monitoring sites in April, June, and July. Samples were collected in dedicated, pre-rinsed one liter plastic bottles and placed in a cooler with ice for transport. The samples were identified and enumerated under a compound microscope immediately upon return to Allied Biological's laboratory. The 2014 microscopic examination data sheets and graphs are provided in the Appendix. In 2014, a single vertical zooplankton tow was conducted at the lake station on each date. The collected sample was preserved in the field, and returned to ABI's lab for analysis.

#### **A PHYTOPLANKTON PRIMER**

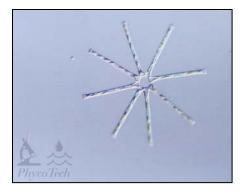
Lakes typically contain three broad categories of phytoplankton (also sometimes referred to as algae). These include filamentous phytoplankton, macroscopic multi-branched phytoplankton (which appear similar to submersed plants), and unicellular phytoplankton. Each category shall be discussed in turn, although the results of the 2014 sampling will focus on the unicellular phytoplankton population.

Filamentous phytoplankton are typically macroscopic (that is, visible with the naked eye), composed of long chains of cells that are attached to a substrate, typically the lake bottom, submersed or emergent vegetation, or rocks. This is called benthic filamentous algae (BFA), and rampant growth can become visible at the surface. As pieces of benthic filamentous algae break apart, it often floats on the surface as dense unsightly mats called floating filamentous algae (FFA). Typically, genera of green algae or blue-green algae develop into nuisance filamentous mats. Abundant nuisance growth of filamentous phytoplankton creates numerous negative impacts to a lake. These can include a decrease in aesthetics, a decrease in recreational uses, increased fishing frustration, and water quality degradation.

Macroscopic multi-branched phytoplankton appears to be submersed plants, especially when viewed in the water column. Physical examination reveals simple structures, no conductive

tissue, and a lack of roots (instead having simplified rhizoids). Although typically only reaching heights of a few inches, under ideal conditions, this type of phytoplankton can reach lengths of several feet, and create a dense carpet on the bottom of a lake. Therefore, it typically does not reach nuisance levels in a lake, save for high use areas such as beaches and other popular swim areas. Since this phytoplankton occupies a similar ecological niche as submersed plants, it's often included in detailed and visual aquatic plant surveys. It provides numerous benefits to a lake system, including sediment stabilization, acting as a nutrient sink, providing invertebrate and fish shelter and habitat, and is one of the first to re-colonize a disturbed area. In the Northeast, muskgrass (*Chara* sp.) and stonewort (*Nitella* sp.) are two of the most common macroscopic multi-branched phytoplankton.

Unicellular phytoplankton are typically microscopic, and consist of individual cells or colonies of cells suspended in the water column. At high enough densities (often called a bloom), they can impart a green or brown (and sometimes, even red) tint to the water column. Unicellular phytoplankton belongs to several taxonomic groups with density and diversity of these groups often varying due to seasonality. When unicellular phytoplankton density becomes elevated it can reduce water clarity (giving the water a "pea soup" appearance), and impart undesirable odors. Usually blue-green algae are responsible for these odors, but other groups or extremely elevated densities can impart them as well. In addition to decreased aesthetics, unicellular phytoplankton blooms can cause degradation of water quality, increase the water temperature (turbid water warms faster than clear water), and can possibly produce a variety of toxins (in the case of blue-green algae), depending on the type of genera present and environmental conditions. Numerous groups of unicellular phytoplankton are common in the Northeast, including diatoms, golden algae, green algae, blue-green algae, euglenoids and dinoflagellates. Each group shall be discussed in turn.



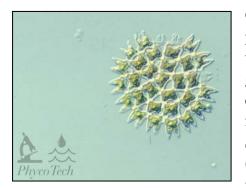
**Diatoms** are ubiquitous as a group, and often possess a rigid silica shell with ornate cell wall markings or etchings. The silica shells settle to the bottom substrate after they die, and under ideal conditions can become stratified. Limnologists can then study historical (and possibly even ancient) population characteristics of diatoms. Some are round and cylindrical (centric) in shape, while others are long and wing-shaped (pennales). They are usually brown in color, and reach maximum abundance in colder or acidic water.

Therefore, they tend to dominate in winter and early spring. Common diatoms in the Northeast include *Fragilaria*, *Cyclotella*, *Navicula*, and *Asterionella* (pictured).



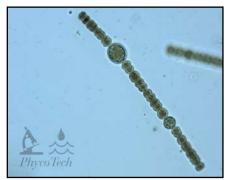
**Golden Algae** are typically yellow or light brown in color. Cell size is usually small oval shaped with a partially empty area, but several genera create colonies of smaller cells. Most have two flagella, and some type of scales or a rigid coating that grants it a fuzzy appearance. However, a few filamentous forms are possible as well. They typically prefer cooler water, so they dominate in the late fall, winter, or early spring. They also tend to bloom at deeper (cooler)

depths. They are common in low nutrient water, and numerous forms produce taste and odor compounds. Common golden algae in the Northeast include *Dinobryon* (pictured), *Mallomonas*, and *Synura*.



**Green Algae** are a very diverse group of unicellular phytoplankton. There is tremendous variability in this group which varies from family to family and sometimes even genus to genus. There are flagellated single cells, multi-cell colonies (some motile), filamentous forms and attached forms, typically with distinct cell shapes light green in color. Some prefer acidic waters, and others highly eutrophic (sewage) conditions. A green algae bloom usually occurs in water with high nitrogen levels. Green algae typically

dominate in mid to late summer in the Northeast. Common genera include *Chlorella*, *Scenedesmus*, *Spirogyra* and *Pediastrum* (pictured).



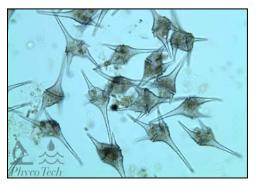
**Blue-green algae** are actually photosynthetic bacteria. Therefore, they tend to be small, simple in structure and lacking interior cell details. Blue-green algae are typically encased in a mucilaginous outer layer. Some genera are adorned with heterocysts, swollen structures capable of fixing nitrogen, a competitive advantage. These types tend to bloom in nitrogen-poor or eutrophic systems. Yet, blue-green algae are tolerant of a wide variety of water chemistries, and boast many oligotrophic forms as well. Blue-green algae

often have gas vesicles which provide increased buoyancy another competitive advantage over other groups of phytoplankton, due to their propensity to shade out others by blooming at the surface. Numerous blue-green algae are documented taste and odor (T&O) producers, and under certain environmental conditions and high enough densities, can produce toxins dangerous to fish, livestock, and possibly humans. Blue-green algae typically dominate a lake system in late summer to early fall. Common blue-green algae that occur in the Northeast include *Anabaena* (pictured), *Aphanizomenon, Microcystis* and *Coelosphaerium*.



**Euglenoids** are typically motile with 0 to 3 (typically 2) flagella, one of which is longer. Euglenoids has plasticity of shape, and usually are grass green in color (although sometime they are clear or even red). Most forms have a distinct red "eyespot. They are often associated with high organic content water, and eutrophic conditions. Common euglenoids that occur in the Northeast include *Euglena* 

(pictured), Phacus, and Trachelomonas.



**Dinoflagellates** are very common in marine environments, in which they often cause toxic blooms. However, toxin production in freshwater genera is very rare. Dinoflagellates are typically single ovoid to spherical cells, but large compared to phytoplankton from other groups. They usually possess two flagella (one wrapped around the middle of the cell) which grant them rotation while they move through the water column. Cellulose plates (armored dinoflagellates) are more

common, but genera without cellulose plates (naked dinoflagellates) also occur. They generally prefer organic-rich or acidic waters, and can impart a coffee-like brown tint to the water at high enough densities. Common dinoflagellates in the Northeast include *Ceratium* (pictured) and *Peridinium*.

#### **PHYTOPLANKTON RESULTS**

In April, the phytoplankton density was considered low, with a favorable community of mostly diatoms, green algae and euglenoids, with traces of golden algae and dinoflagellates at the inlet station only. Diversity was high at the inlet station with eleven genera, and moderate at the lake station with eight total genera.

Algal Group	Inlet Station			
% Abundance	4/28/2014	6/6/2014	7/30/2014	
Diatoms	33.3%	7.7%	7.7%	
Golden Algae	3.3%	69.2%		
Protozoa		15.4%		
Green Algae	26.8%	7.7%	69.2%	
Blue-green Algae				
Dinoflagellates	3.3%		23.1%	
Euglenoids	33.3%			
Total Orgs / mL	300	130	130	

 Table 10. Inlet Station

Algal Group			
% Abundance	4/28/2014	6/6/2014	7/30/2014
Diatoms	50.0%	16.6%	
Golden Algae		77.8%	13.6%
Protozoa		5.6%	
Green Algae	30.8%		13.6%
Blue-green Algae			9.1%
Dinoflagellates			63.6%
Euglenoids	19.2%		
Total Orgs. / mL	260	180	220

Table 11. Lake Station

improved to 8.5 feet.

By June, the phytoplankton density and diversity decreased at both sites, with assemblages comprised of five genera at the inlet station, and only three genera at the lake station. Each of the phytoplankton assemblages consisted of trace amounts of diatoms, golden algae, and protozoa, while the inlet station also contained a trace of green algae. Water clarity at this time was

On the final sampling date on July 30<sup>th</sup>, the algal community at each sampling site remained at light density, with 130 organisms per liter at the inlet, and 220 organisms per liter at the lake station. The assemblage at the inlet site was dominated by five genera of green algae accounting for 69.2 percent of population. A light density bloom of *Peridinium* (140 orgs./mL), a dinoflagellate, was responsible for 63.6 percent of the assemblage at the lake station, with additionally, traces of golden, green and blue-green algae.

#### **A ZOOPLANKTON PRIMER**

Zooplankton provides an important link in a typical lake's food web between phytoplankton and developing/juvenile stages of fish. In general, zooplankton feed on phytoplankton, while fish in turn feed on zooplankton. The rate of phytoplankton feeding efficiency is primarily based on body size, but zooplankton group, and to some effect specific genera, also plays an important role. There are three main groups of zooplankton found in freshwater systems: rotifers, cladocera, and copepods.



Rotifers are a diverse group of zooplankton, very common in lakes and marine environments alike. Rotifers are generally the smallest zooplankton of the three groups, and thus typically the least efficient phytoplankton grazers. Feeding preferences are determined primarily by mouth structures, and include generalist feeders (omnivores), which eat any small organic detritus encountered, and predators, which eat other smaller rotifers and small phytoplankton. Generalist feeders include *Filinia, Keratella, Lecane, Euchlanis*, and *Brachionus*. Predator genera include *Polyarthra* (larger species), *Asplanchna, Synchaeta*, and

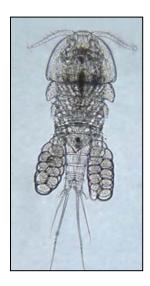
Trichocerca.

Cladocera are less diverse, but also very common in freshwater lakes. They are sometimes called "water fleas". They spend most of their lifecycle reproducing via parthenogenesis (asexual reproduction with an all female population) only less efficient sexual reproduction when switching to environmental conditions decline. Some genera (such as Daphnia) can be quite large (up to 5.0 mm long, visible without magnification), and thus can be classified as highly Most cladocera efficient phytoplankton grazers. are phytoplankton grazers, although their diet includes most organic matter ingested, including bacteria and protozoa. Body



size

(and thus mouth size) determines feeding efficiency, but ironically the larger-bodied genera are easier to see by predaceous fish, and thus typically have reduced numbers in populations of zooplanktivorous fish. *Daphnia* are the most efficient phytoplankton feeders, while *Ceriodaphnia, Bosmina* and *Eubosmina* are less efficient. There are a few predator genera as well, including *Polyphemus* and *Leptodora*.



Copepods are almost excusive to freshwater lake systems (not streams or rivers) and estuarine and marine systems. Of the six suborders native to the United States, three are parasitic, and three are free living. One of the free living, *Harpacticoida* are exclusively benthic and thus often not collected in traditional plankton tows (unless the bottom sediments are disturbed). The remaining two suborders, the Calanoida and the Cyclopoida are of primary concern during lake studies. All copepods have several naupilar stages, followed by several immature stages, before reaching an adult stage. Both suborder adults are considered large bodied zooplankton, but have distinct feeding preferences. Calanoids are almost exclusively phytoplankton feeders and have even demonstrated selective feeding strategies. Cyclopoids have mouth parts suitable for biting and seizing prey. Their diet is primarily other crustacean zooplankton

(including cannibalism on younger life stages), as well as phytoplankton and organic detritus ingestion, but less efficiently.

Zooplankton samples were collected with an 80 um Nitex plankton net. At the Lake Station, a single vertical tow was performed to a depth of 18 feet. Using as little site water as possible, the sides of the net were rinsed of any trapped zooplankton, concentrating the organisms into the net bottom. This concentrate was then emptied into a clean 1000 mL HDPE sample bottle. Immediately after collection, the sample was preserved with an equal amount of 10% sucrose formalin, to achieve a 5% solution. Sucrose was added to the preservative to help maintain carapace integrity. The samples were then placed in a cooler stocked with blue ice. On arrival at ABI's laboratory, the samples were stored in a dark refrigerator until the samples were identified and enumerated.

In the laboratory, each sample was manually mixed for about one minute, before a one mL subsample was removed using a calibrated syringe. The subsample was placed on a Sedgewick-Rafter counting cell, and examined under a compound microscope at 100X magnification. By using calibrated guides on the microscope stage, the entire one mL sample was examined, and any zooplankton were identified and enumerated to the lowest practical taxa using regionally appropriate taxonomic keys. This procedure was repeated two more times to generate three replicate counts. The counts were then averaged, and back-calculated to achieve an organism per liter density. The zooplankton count data sheets in the Appendix describe the step by step procedures for all three replicates, and the final averaged densities. Also included in the Appendix are pie charts depicting the sample date zooplankton group distribution.

Zooplankton Group	4/28/2014	6/30/2014	7/30/2014
Rotifers	94.3%	86.1%	86.05%
Cladocera	1.2%	12.5%	9.05%
Copepoda	4.5%	1.4%	4.9%
Total Zooplankton (Orgs. / mL)	1285	7079	1391

#### 2014 Zooplankton Results

Table 12. Mount Kemble Lake 2014 Zooplankton Group Percent Abundance Distribution

In April, overall zooplankton density was 1,285 organisms per liter, which is considered high. Sample diversity was moderate with seven different genera observed. At this time Rotifers accounted for 94.3 percent of the total zooplankton community with *Keratella*, accounting individually for 82.6 percent. Additionally, a single genus of Cladoceran (*Daphnia*) and Copepod (*Cyclopoid nauplii*) were also represented in the zooplankton community, with *Daphnia* individually only accounting for 1.2 percent of the community.

The June sampling revealed a higher diversity rotifers, with nine genera accounting for 6,095 organisms per liter, which is approximately six times greater than any observed rotifer density over the past two seasons. *Keratella* accounted for 49.4 percent of the rotifer population, and 42.5 percent of the total zooplankton community. The second most abundant genera of rotifer, *Polyarthra*, represented 28.9 percent of the rotifers, and 24.8 percent of the total community. *Daphnia* density also increased significantly from the previous sampling, with the individual population increasing from 15 organisms per liter in April to 882 organisms per liter at this time. Additionally, a single genus of Copepod (*Cyclopoid nauplii*) was also observed, although representing only 1.5 percent of the zooplankton community.

On the final sampling date of July 30<sup>th</sup>, the zooplankton composition although still diverse, was more representative of the April sampling, with a total population density of 1,391 organisms per liter, colonized by nine genera of rotifers. Also at this time there were no observed genera of *Daphnia* within the collected sample. Overall similar to the previous two seasons, rotifers, more specifically *Keratella* and *Polyarthra* were the most dominant genera of zooplankton during the season. The low density of *Daphnia* is likely a result of the low densities of observed phytoplankton growth, as the population dynamics of *Daphnia* are generally directed by the intensity of phytoplankton growth. It is more concerning from a fisheries management perspective, as *Daphnia* provide the base of the fish population food chain.

#### DISCUSSION

The 2015 management program of Mount Kemble Lake will continue to focus on the control of nuisance densities of plant and algae growth. The target aquatic macrophyte species observed at Mount Kemble Lake in 2014 were southern naiad, curly-leaf and leafy pondweed. It is recommended that localized applications of the contact aquatic herbicide Reward continue to be utilized through the season for its ability to selectively control nuisance submerged vegetation by rapid absorption into the target plant. In addition, it is beneficial to allow certain amounts of plants to persist in the lake to provide dissolved oxygen, habitat and compete for nutrients required for nuisance plant and algae development. The growth of leafy pondweed (*Potamogeton foliosus*) and southern naiad (*Najas guadalupensis*) should be encouraged in areas of the lake, such as the northern inlet, where they are not interrupting recreational activities or reducing the aesthetic appeal of the lake.

Copper sulfate will continue to provide the most effective and cost efficient management method for controlling nuisance density filamentous and planktonic algae growth. It should be reminded that Copper Sulfate has acknowledged negative impacts on zooplankton populations, with localized targeted applications recommended for only nuisance growth of filamentous algae, and limited use on planktonic algae blooms only at times when water clarity is significantly impaired. Numerous other copper and non-copper based algaecides are available and at the request of the Association, ABI would be happy to discuss these alternatives. The success of the application of Aluminum Sulfate on July 23<sup>rd</sup> was evident during the July 30<sup>th</sup> water quality sampling, as water clarity increased approximately three feet, and concentrations of nitrate and total phosphorous were significantly reduced. Based on the observed results of this application, it is recommended that an application of Aluminum Sulfate be performed during the month of May in 2015 to reduce nutrient levels in the water column prior to the height of the summer aquatic plant and algae growth period, and compare results with those observed from the 2014 application.

The current Mount Kemble Lake Water Quality Monitoring Program is well-designed, and provides suitable water quality data allowing for proactive management of the lakes' environment and reduces the opportunity for the development of problematic situations. It is important to continue water quality monitoring on a regular yearly basis over the long-term to build a baseline data record which will assist biologists in developing more quantitative analysis for greatest possible management procedures.

The entire staff at Allied Biological appreciates the opportunity to be of service to the Mount Kemble Lake Association and looks forward to assisting the Association on the stewardship of Kemble Lake for many years to come.

Sincerely,

Robert Schindler **Bob Schindler Aquatic Biologist** 

# **APPENDIX**

APPENDIX A: WATER QUALITY PARAMETER DESCRIPTION APPENDIX B: AQUATIC MACROPHYTE GUIDE APPENDIX C: WATER QUALITY SAMPLING MAP APPENDIX D: PHYTOPLANKTON ENUMERATION CHARTS APPENDIX E: ZOOPLANKTON ENUMERATION CHARTS APPENDIX F: DISSOLVED OXYGEN – TEMP. PROFILES APPENDIX G: LAB DATA REPORTS

# **APPENDIX A: WATER QUALITY PARAMETER DESCRIPTIONS**

#### Temperature

Temperature is measured in degrees Celsius, and is very important to aquatic biota. Several factors affect temperature in a lake system, including air temperature, season, wind, water flow through the system, and shade trees. Turbidity can also increase water temperature as suspended particles absorb sun rays more efficiently. Water depth also affects temperature. In general, deeper water remains cooler during the summer months.

Temperature preferences vary among aquatic biota. Since water temperature typically varies between 5 °C and 30 °C during the season, most aquatic biota can flourish under this wide range of temperatures. Of more concern is thermal shock, which occurs when temperature rapidly changes in a short amount of time. Some aquatic biota can become stressed when temperature changes as little as 1-2 °C in a 24 hour period.

#### **Dissolved Oxygen**

Dissolved Oxygen is the measurement of the amount of oxygen freely available to aquatic biota in water. Several factors play a role in affecting the amount of dissolved oxygen in the water. These factors include temperature (warmer water holds less dissolved oxygen), low atmospheric pressure (such as higher altitude) decreases the solubility of oxygen, mineral content of the water can reduce the water's dissolved oxygen capacity, and water mixing (via wind, flow over rocks, or thermal upwelling) increases dissolved oxygen in the water. In addition, an over abundance of organic matter, such as dead algae or plants causes rapid aerobic bacteria growth. During this growth, bacteria consume oxygen during respiration, which can cause the water's dissolved oxygen to decrease.

Dissolved oxygen has a wide range, from 0 mg/L to 20 mg/L. To support diverse aquatic biota, 5-6 mg/L is minimally required, but 9-10 mg/L is an indicator of better overall water quality. Dissolved oxygen reading of below 4 mg/L is stressful to most aquatic organisms, especially fish.

#### Water Clarity

Transparency (or visibility) is measured with a Secchi disc, and can provide an experienced biologist with a quick determination of a lake's water quality. In short, higher visibility indicates a cleaner (and healthier) aquatic system. Cloudy conditions could indicate nutrient rich sediments entering the lake or excessive algal blooms due to nutrient availability, leading to a degradation of water quality. Clear conditions allow greater light penetration and the establishment of a deeper photic zone. The photic zone is the depth of active photosynthesis carried out by plants and algae. A byproduct of photosynthesis is dissolved oxygen, required for use by higher aquatic organisms, such as zooplankton and fish.



#### **Total Hardness**

Hardness is a measure of dissolved salts in the water, usually calcium, but also magnesium and iron. Hardness is usually influenced by the rock and soil types of the watershed, and the amount of runoff over these surfaces. Hardness can be measured for only calcium content (Hardness (Ca)), or for all three salts, called Total Hardness. Water with Hardness (Ca) less than 10 mg/L can only support sparse aquatic biota. Freshwater typically has a Hardness (Ca) level from 4 to 100 mg/L. In general, the degree of total hardness can be classified according to the table to the right.

## Alkalinity

Alkalinity is the measure of the water's capacity to neutralize acids. A higher alkalinity can buffer the water against rapid pH changes, which in turn prevents undue stress on aquatic biota due to fluctuating pH levels. The alkalinity of a lake is primarily a function of the watersheds soil and rock composition. Limestone, dolomite and calcite are all a source of alkalinity. High levels of precipitation in a short amount of time can decrease the waters alkalinity. A typical freshwater lake has an alkalinity of 20-200 mg/L. A lake with a low alkalinity typically also has a low pH, which can limit the diversity of aquatic biota.

## pН

The measurement of acidity or alkalinity of the water is called pH (the "potential for hydrogen"). Several factors can impact the pH of a lake, including precipitation in a short amount of time, rock and soil composition of the watershed, algal blooms (increase the pH), and aquatic plant decomposition (decreases the pH). A pH level of 6.5 to 7.5 is considered excellent, but most lake systems fall in the range of 6.0 to 8.5. Aquatic biota can become stressed if the pH drops below 6.0, or increases above 8.5 for an extended amount of time.

Most aquatic biota are adapted to specific pH ranges. When the pH fluctuates rapidly, it can cause changes in aquatic biota diversity. Immature stages of aquatic insects and juvenile fish are more sensitive to low pH values than their adult counterparts. Therefore, a low pH can actually inhibit the hatch rate and early development of these organisms.

## Conductivity

Conductivity is the measure of water's ability to conduct an electrical current, and is measured in umhos/cm, the higher the number of charged particles(ions) in the water, the easier for electricity to pass through it. Conductivity is useful in lake management by estimating the dissolved ionic matter in the water, the lower the conductivity, the higher the quality of water (oligotrophic). A higher conductivity usually indicates an abundance of plant nutrients (total phosphorous and nitrate), or eutrophic conditions. Industrial discharge, road salt runoff, and septic tank leaching can increase conductivity. Distilled water has a conductivity of 0.5 to 2.0 umhos/cm, while drinking water conductivity typically ranges from 50to 1,500 umhos/cm. Conductivity below 500 umhos/cm is considered ideal in a lake system.

#### Nitrate

Nitrates are chemical compounds derived from nitrogen and oxygen. Nitrogen is needed by all plants and animals to make proteins needed for growth and reproduction. Nitrates are generated during plant and animal decomposition, from man-made sources, and from livestock and waterfowl sources. Man-made sources of nitrates include septic system leaching, fertilizer runoff, and improperly treated wastewater. Freshwater lake systems can potentially receive large nitrate inputs from waterfowl, specifically large flocks of Canada geese. An increase in nitrate levels can in turn cause an increase in total phosphorous levels. A nitrate level greater than 0.3 mg/L can promote excessive growth of aquatic plants and algae.

#### **Total Phosphorous**

Total phosphorous is a chemical compound derived from phosphorous and oxygen. Total phosphorous is usually present in freshwater in low concentrations, and is often the limiting nutrient to aquatic plant growth. However, man-made sources of phosphorous include septic system leaching, fertilizer runoff, and improperly treated wastewater. These phosphorous inputs usually enter a freshwater lake system during rain events, and bank erosion.

A total phosphorous level greater than 0.03 mg/L can promote excessive aquatic plant growth and decomposition, either in the form of algal blooms, or nuisance quantities of aquatic plants. This process is called eutrophication, and when induced or sped up by man-made nutrient inputs, it is called cultural eutrophication. As a result of this excessive growth, recreational activities, such as swimming, boating, and fishing in the lake can be negatively impacted. In addition, aerobic bacteria will thrive under these conditions, causing a decrease in dissolved oxygen levels which can negatively impact aquatic biota such as fish.

#### **Total Suspended Solids**

Total suspended solids refer to all of the particulate matter suspended in the water column. When these solids settle to the bottom of a water body (a process called siltation), they become sediments. There are two components that make up total suspended solids: inorganic and organic. The inorganic portion is usually considerably higher than the organic portion and includes silts, clays, and soils. Organic solids include algae, zooplankton, bacteria and organic debris. All these solids create turbid (or "muddy") conditions. The geology and vegetation of a watershed affect the amount of suspended solids that enter a lake system. Most suspended solids originate from accelerated soil erosion from agricultural operations, logging activities, and construction activities. Another source is the disturbance of bottom sediments from dredging activities, grazing of bottom feeding fish, and recreational activities such as boating.

#### Ammonia

Ammonia is a type of nitrogen compound used by plants and algae to support growth. Ammonia content in a body of water is influenced by decaying plants and animals, animal waste, industrial waste effluent, agricultural runoff, and atmospheric nitrogen gas transfer. A concentration exceeding 0.30 mg/L can promote excessive plant and algae growth. Elevated ammonia levels can increase nitrification, which in turn depletes the oxygen content of water. Extremely high ammonia levels can be toxic to aquatic biota (such as fish).

#### **APPENDIX B: AQUATIC MACROPHYTE GUIDE**

**Small Duckweed** (*Lemna minor*. Common Names: Small duckweed, water lentil, lesser duckweed. **Native.**). Small duckweed is a free floating plant, with round to oval-shaped leaf bodies typically referred to as fronds. The fronds are small (typically less than 0.5 cm in diameter), and it can occur in large densities that can create a dense mat on the water's surface. Each frond contains three faint nerves, a single root (a characteristic used to distinguish it from other duckweeds), and no stem. Although it can produce flowers, it usually reproduces via budding at at a tremendous rate.



Its population can double in three to five days. Since it is free floating, it drifts with the wind or water current, and is often found intermixed with other duckweeds. Since it's not attached to the sediment, it derives nutrients directly from the water, and is often associated with eutrophic conditions. It over winters by producing turions late in the season. Small duckweed is extremely nutritious and can provide up to 90% of the dietary needs for waterfowl. It's also consumed by muskrat, beaver and fish, and dense mats of duckweed can actually inhibit mosquito breeding.



**Curly-leaf Pondweed** (*Potamogeton crispus*. Common Name: curly-leaf pondweed. **Invasive.**): Curly-leaf pondweed has spaghetti-like stems that often reach the surface by mid-June. Its submersed leaves are oblong, and attached directly to the stem in an alternate pattern. The margins of the leaves are wavy and finely serrated, hence its name. No floating leaves are produced. Curly-leaf pondweed can tolerate turbid water conditions better than most other macrophytes. In late summer, Curly-leaf pondweed enters its summer dormancy stage. It naturally dies off (often creating a sudden loss of habitat and releasing nutrients into the water to fuel algae growth) and produces vegetative buds called turions. These turions germinate when the water gets cooler in the autumn and

give way to a winter growth form that allows it to thrive under ice and snow cover, providing habitat for fish and invertebrates.



**Leafy Pondweed** (*Potamogeton foliosus*: Common Name: leafy pondweed. **Native**.): Leafy pondweed has freely branched stems that hold slender submersed leaves that become slightly more narrow as they approach the stem. The leaf contains 3-5 veins and often tapers to a point. No floating leaves are produced. It produces early season fruits in tight clusters on short stalks in the leaf axils. These early season fruits are often the first grazed upon by waterfowl during the season. Muskrat, beaver, deer and even moose also graze on the fruit. It inhabits

a wide range of habitats, but usually prefers shallow water. It has a high tolerance for eutrophic conditions, allowing it to even colonize secondary water treatment ponds.

Southern Naiad (Najas guadalupensis. Common Names: Southern water nymph, bushy pondweed. Native.): Southern naiad is an annual aquatic plant that can form dense stands of rooted vegetation. Its ribbon-like leaves are dark-green to greenish-purple, and are wider and less pointed than slender naiad. Flowers occur at the base of the leaves, but are so small, they usually require magnification to detect. Southern naiad is widely distributed, but is less common than slender naiad in northern zones. Southern naiad reproduces seeds by and fragmentation.





**Variable-leaf Pondweed** (*Potamogeton diversifolius*. Common Names: Water-thread pondweed, variable-leaf pondweed, snailseed pondweed. **Native**.): Variable-leaf pondweed have freely-branched stems emerging from slender rhizomes. The submersed leaves are narrow and linear with one obvious midvein bordered by a row of hollow cells. The floating leaves are shaped like an ellipse, but are usually less than 4 cm long, Variable-leaf pondweed fruit spikes are

produced in two distinct forms. It occurs in lakes, ponds, rivers and streams and prefers soft sediment and water less than 2 meters deep. Waterfowl graze on the fruit, and local fauna often graze on the stems and leaves.

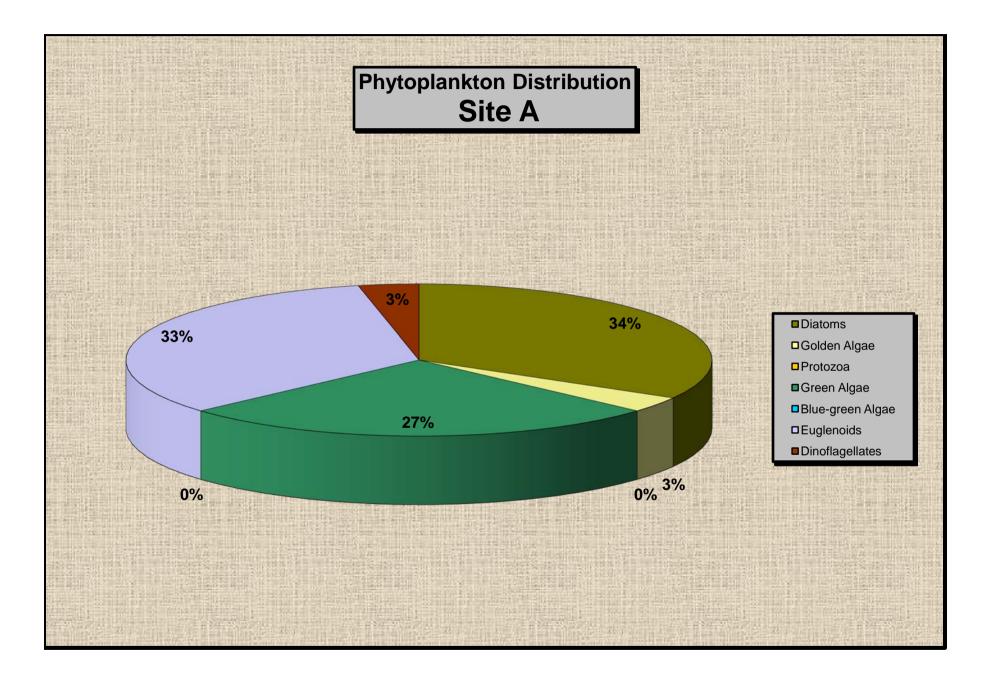
# North Inlet WO Station Lake Station (Bottom WQ Station-Site #5) Located in ~30 feet of water (deepest site located) © 2013 Google Google earth **Mount Kemble Lake** 2014 Water Quality Sampling Sites Map Morris County, New Jersey $\circ$ = North Inlet WQ Site • = Lake Station WQ Site Allied Biological 580 Rockport Road Hackettstown, NJ 07840 (908) 850-0303 FAX 850-4994 1.5

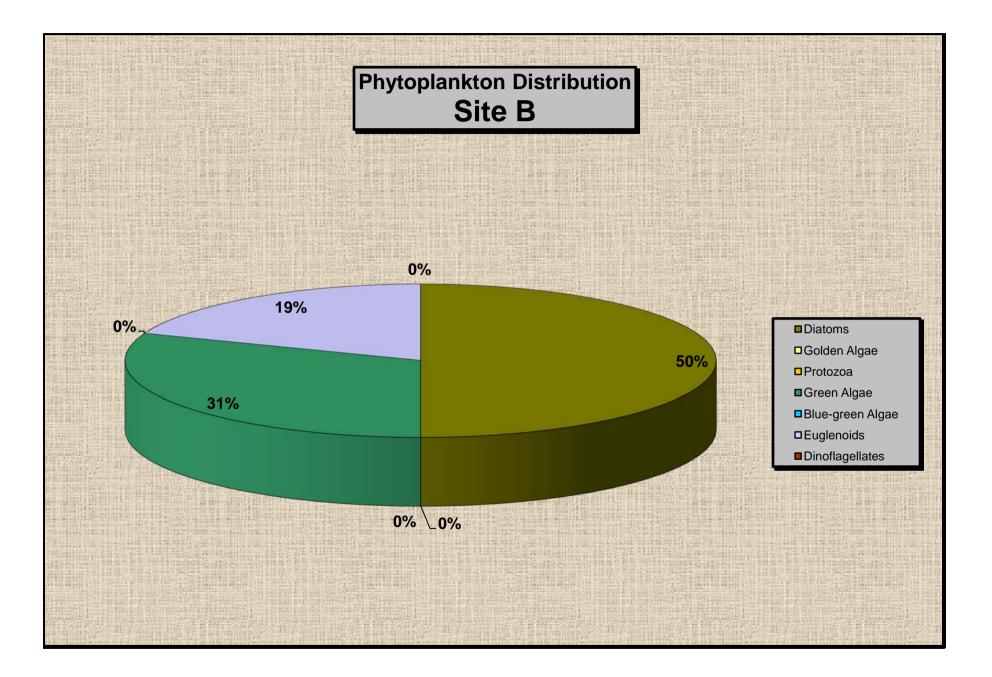
# Mount Kemble Lake 2014 Water Quality Monitoring Sites



# **MICROSCOPIC EXAMINATION OF WATER**

Sample from			mble								
Collection Date: 4	/28/2014	4		Examination Date	e: 4/30/2	014		Amount Examine	ed: 200	ml.	
Site A: North Sta	tion (Inl	let)		Site B: Lake Stati	on			Site C:			
BACILLARIOPHYTA (Diatoms)	А	В	с	CHLOROPHYTA (Green Algae)	А	в	с	CYANOPHYTA (Blue-green Algae)	А	в	с
Asterionella	40	40		Actinastrum				Anabaena			
Cocconeis	-	-		Ankistrodesmus				Anacystis			
Cyclotella	20	50		Chlamydomonas				Aphanizomenon			
Cymbella				Chlorella	10	60		Coelosphaerium			
Diatoma				Chlorococcum				, Cylindrospermum			
Fragilaria		20		Closterium				Gomphosphseria			
Melosira				Coelastrum				Lyngbya			
Meridion				Cosmarium				Microcystis			
Navicula	20			Desmodium				Nostoc			
Pinnularia				Eudorina				Oscillatoria			
Rhizosolenia				Gloeocystis				Pseudoanabaena			
Stephanodiscus				Micrasterias				Synechocystis			
, Synedra		20		Micratinium				, , ,			
Tabellaria	20			Microspora							
Stauroneis				Mougeotia				Total Blue-green Algae	0	0	0
Nitzschia				Oedogonium				EUGLENOPHYTA			
Denticula				Oocystis				(Euglenoids)	Α	в	с
Total Diatoms	100	130	0	Pandorina				Euglena	70	30	
CHRYSOPHYTA				Pediastrum				Lepocinclis			
(Golden Algae)	Α	в	с	Phytoconis				Phacus			
Dinobryon	10			Rhizoclonium				Trachelomonas	30	20	
Mallomonas				Scenedesmus	10	20					
Synura				Sphaerocystis	60						
Tribonema				Spirogyra							
Uroglenopsis				Staurastrum							
Vaucheria				Tetraedron				Total Euglenoids	100	50	0
				Ulothrix				PYRRHOPHYTA			
				Volvox				(Dinoflagellates)	Α	в	с
				Palmella				Ceratium	10		
Total Golden Algae	10	0	0	Euastrum				Peridinium			
PROTOZOA				Zygnema				Gymnodinium			
	Α	В	С	Dictyosphaerium							
Actinophrys				Quadrigula							
Vorticella				Bottryococcus							
				Trackdomuccs							
				Eaestrum							
Total Protozoa	0	0	0	Total Green Algae	80	80	0	Total Dinoflagellates	10	0	0
SITE	Α	В	С	moderate-high sample div	versity. The	assemblage	e at both S	density at both sites is light tations was a mixture of dia	toms, gree	n algae and	l
TOTAL GENERA:	11	8		0	0	dinoflagellat	es were ob	served at the North Station	only. Wate	er clarity was	3
TRANSPARENCY:	4.0'	4.0'		considered fair on this dat	ie.						
ORGANISMS PER MILLILITER:	300	260									

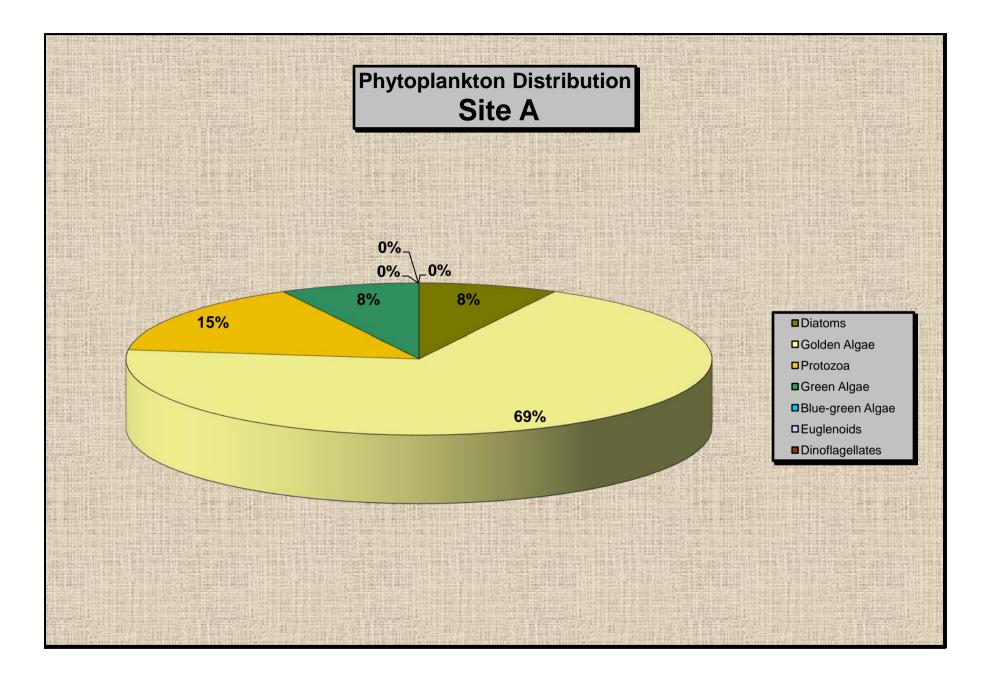


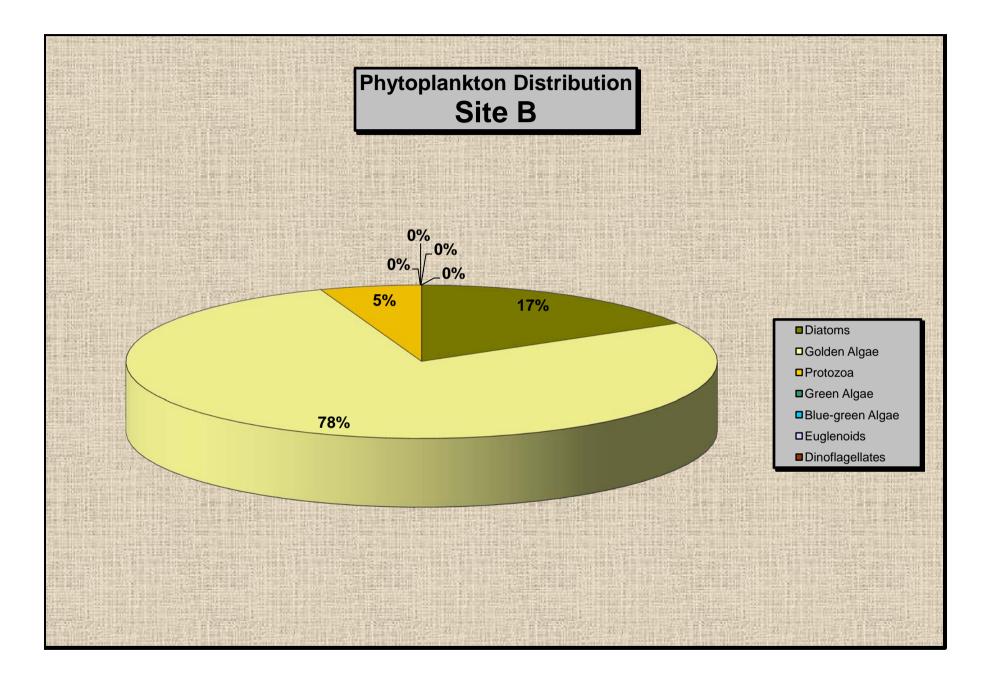




# **MICROSCOPIC EXAMINATION OF WATER**

		141									
Sample from		unt Ke	mble								
Collection Date: 6				Examination Date		14		Amount Examine	<b>d:</b> 200	ml.	
Site A : North Sta	ntion (Inl	let)		Site B: Lake Stati	on			Site C:			
BACILLARIOPHYTA (Diatoms)	А	В	с	CHLOROPHYTA (Green Algae)	А	в	с	CYANOPHYTA (Blue-green Algae)	А	в	с
Asterionella	10	30		Actinastrum				Anabaena			
Cocconeis				Ankistrodesmus				Anacystis			
Cyclotella				Chlamydomonas				Aphanizomenon			
Cymbella				Chlorella				, Coelosphaerium			
Diatoma				Chlorococcum				, Cylindrospermum			
Fragilaria				Closterium	10			Gomphosphseria			
Melosira				Coelastrum				Lyngbya			
Meridion				Cosmarium				Microcystis			
Navicula				Desmodium				Nostoc			
Pinnularia				Eudorina				Oscillatoria			
Rhizosolenia				Gloeocystis				Pseudoanabaena			
Stephanodiscus				Micrasterias				Synechocystis			
Synedra				Micratinium				, , ,			
Tabellaria				Microspora							
Stauroneis				Mougeotia				Total Blue-green Algae	0	0	0
Nitzschia				Oedogonium				EUGLENOPHYTA			
Denticula				Oocystis				(Euglenoids)	Α	в	с
Total Diatoms	10	30	0	Pandorina				Euglena			
CHRYSOPHYTA				Pediastrum				Lepocinclis			
(Golden Algae)	А	в	с	Phytoconis				Phacus			
Dinobryon	20	140		Rhizoclonium				Trachelomonas			
Mallomonas	70			Scenedesmus							
Synura	-			Sphaerocystis							
Tribonema				Spirogyra							
Uroglenopsis				Staurastrum							
Vaucheria				Tetraedron				Total Euglenoids	0	0	0
				Ulothrix				PYRRHOPHYTA			
				Volvox				(Dinoflagellates)	А	в	с
				Palmella				Ceratium			
Total Golden Algae	90	140	0	Euastrum				Peridinium			
PROTOZOA				Zygnema				Gymnodinium			
	А	в	с	Dictyosphaerium							
Actinophrys	20	10		Quadrigula							
Vorticella	-	-		Bottryococcus							
	1		İ	Trackdomuccs	1	1	1			1	i
				Eaestrum							
Total Protozoa	20	10	0	Total Green Algae	10	0	0	Total Dinoflagellates	0	0	0
SITE	A	В	C	NOTES: Algal density dec	reased and			ht and favorable at sites A a ted by golden algae. Trace			
TOTAL GENERA:	5	3						increased and is now cons			
TRANSPARENCY:	8' est.	8.5'		sites.			,		5		
ORGANISMS PER MILLILITER:	130	180		1							

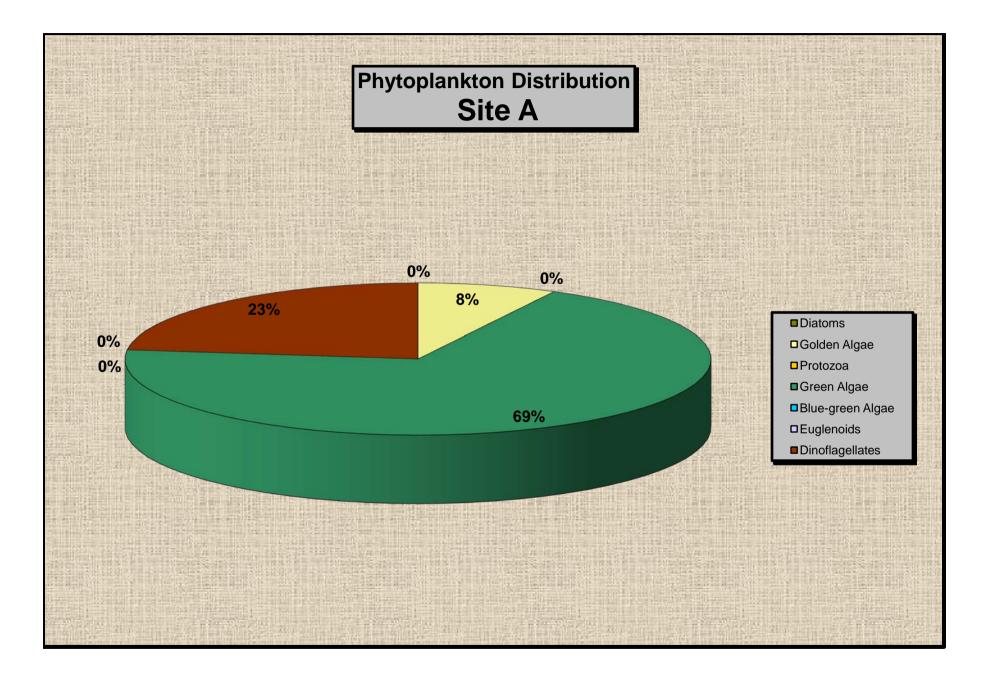


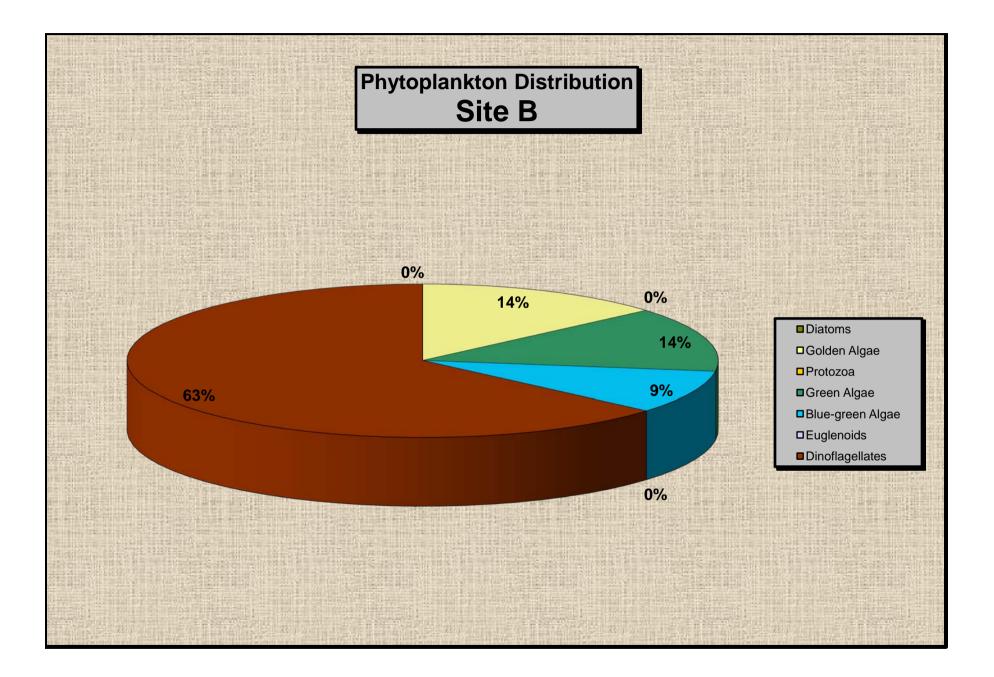




# **MICROSCOPIC EXAMINATION OF WATER**

Sample from	<b>։</b> : Moւ	unt Ke	emble								
Collection Date: 7				Examination Date		014		Amount Examine	ed: 200	ml.	
Site A: North Sta	ition (Inl	et)		Site B: Lake Stati	on			Site C:			
BACILLARIOPHYTA (Diatoms)	А	В	с	CHLOROPHYTA (Green Algae)	А	в	с	CYANOPHYTA (Blue-green Algae)	А	в	с
Asterionella				Actinastrum				Anabaena			
Cocconeis				Ankistrodesmus			1	Anacystis			
Cvclotella				Chlamydomonas			1	Aphanizomenon		20	
Cymbella				Chlorella				Coelosphaerium			
Diatoma				Chlorococcum				, Cylindrospermum			
Fragilaria				Closterium	10	10		Gomphosphseria			
Melosira				Coelastrum	50			Lyngbya			
Meridion				Cosmarium				Microcystis			
Navicula				Desmodium				Nostoc			
Pinnularia				Eudorina				Oscillatoria			
Rhizosolenia				Gloeocystis		20		Pseudoanabaena			
Stephanodiscus				Micrasterias				Synechocystis			
Synedra				Micratinium				, , ,			
Tabellaria				Microspora							
Stauroneis				, Mougeotia				Total Blue-green Algae	0	20	0
Nitzschia				Oedogonium	10			EUGLENOPHYTA			
Denticula				Oocystis				(Euglenoids)	А	в	с
Total Diatoms	0	0	0	Pandorina				Euglena			
CHRYSOPHYTA				Pediastrum	10			Lepocinclis			
(Golden Algae)	Α	в	с	Phytoconis				Phacus			
Dinobryon	10	10	Î.	Rhizoclonium				Trachelomonas			
Mallomonas		20		Scenedesmus							
Synura	Î			Sphaerocystis							
Tribonema				Spirogyra							
Uroglenopsis	Î			Staurastrum	10						
Vaucheria	Î			Tetraedron				Total Euglenoids	0	0	0
	1			Ulothrix				PYRRHOPHYTA			
	Î			Volvox				(Dinoflagellates)	Α	в	с
				Palmella				Ceratium	10		
Total Golden Algae	10	30	0	Euastrum				Peridinium	20	140	
PROTOZOA				Zygnema				Gymnodinium			
	Α	В	с	Dictyosphaerium							
Actinophrys	1		I	Quadrigula							
Vorticella				Bottryococcus							
				Trackdomuccs							
				Eaestrum							
Total Protozoa	0	0	0	Total Green Algae	90	30	0	Total Dinoflagellates	30	140	0
SITE	Α	В	С					while site A remained simila al diversity is considered m			Site A
TOTAL GENERA:	8	6		(North) is dominated by th	e green alg	ae called C	oelastrum.	Site B (Lake Station) is dor	minated by	the dinoflag	gellate
TRANSPARENCY:	5' est.	5.0'		observed at site B only. W	ater clarity	decreased		rved this week. Trace amou es since the last sampling e			
ORGANISMS PER MILLILITER:	130	220		considered fair at site B, v	vhile site A	is poor.					





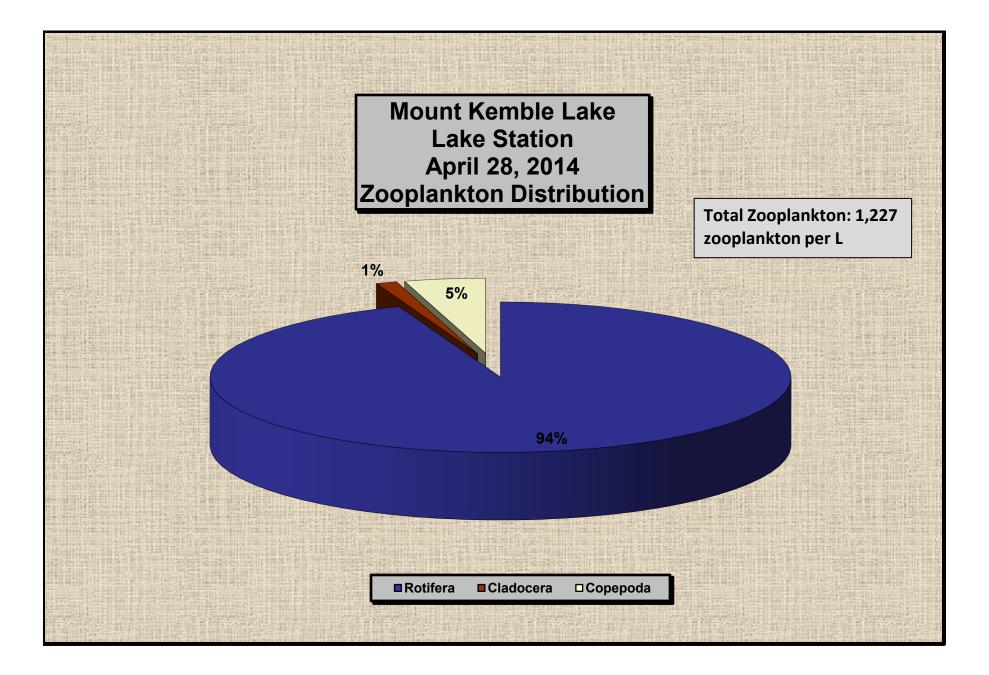
# Zooplankton Count Results

				Replicate	•	Total/3	x1000 mL	Water	# organisms
Order	Family	Genus	А	В	С	(# per mL)	( = 1 L)	sampled (L)	per L
Ploima	Asplanchnidae	Asplanchna	1	1		1	667	68.8	10
	Brachionidae	Brachionous	2	4	3	3	3000	68.8	44
		Keratella	66	94	59	73	73000	68.8	1061
		Kellicottia	1		1	1	667	68.8	10
	Synchaetidae	Polyarthra	5	6	7	6	6000	68.8	87
								Total:	1212
Cladocera									
	Daphnidae	Daphina		3		1.00	1000	68.8	15
								Total:	15
Cyclopoida	Cyclopoidae	Cyclopoid nauplii	6	4	2	4.00	4000	68.8	58
								Total:	58
	Ploima Ploima Cladocera	Ploima Asplanchnidae Brachionidae Synchaetidae Cladocera Daphnidae	Ploima       Asplanchnidae       Asplanchna         Brachionidae       Brachionous         Keratella       Kellicottia         Synchaetidae       Polyarthra         Cladocera       Daphnidae         Daphnidae       Daphina	Ploima       Asplanchnidae       Asplanchna       1         Brachionidae       Brachionous       2         Keratella       66         Keratella       1         Synchaetidae       Polyarthra       5         Daphnidae       Daphina       1	OrderFamilyGenusABPloimaAsplanchnidaeAsplanchna11BrachionidaeBrachionous24Keratella6694Keratella6694Kellicottia11SynchaetidaePolyarthra56CladoceraImage: Company of the second	OrderFamilyGenusABCPloimaAsplanchnidaeAsplanchna111BrachionidaeBrachionous243BrachionidaeBrachionous243Keratella669459Kellicottia111SynchaetidaePolyarthra567CladoceraImage: Company of the synchaetidaeDaphina3Image: Company of the synchaetidaeDaphnidaeDaphina3Image: Company of the synchaetidaeImage: Company of the synchaetidae	OrderFamilyGenusABC(# per mL)PloimaAsplanchnidaeAsplanchna1111BrachionidaeBrachionous2433Image: Second Secon	OrderFamilyGenusABC(# per mL)(=1 L)PloimaAsplanchnidaeAsplanchna111667PloimaAsplanchnidaeAsplanchna111667BrachionidaeBrachionous24333000CarlosKeratella666945997373000CarlosKellicottia111667SynchaetidaePolyarthra56766000CladoceraImage: CarlosImage: CarlosImage: CarlosImage: CarlosImage: CarlosImage: CarlosDaphnidaeDaphinaImage: CarlosImage: CarlosImage	Order         Family         Genus         A         B         C         (# per mL)         (= 1 L)         sampled (L)           Ploima         Asplanchnidae         Asplanchna         1         1         667         68.8           Ploima         Asplanchnidae         Asplanchna         1         1         1         667         68.8           Brachionidae         Brachionous         2         4         3         3         3000         68.8           Image: Synchaetidae         Brachionous         2         4         3         3         3000         68.8           Image: Synchaetidae         Keratella         666         94         59         73         73000         68.8           Image: Synchaetidae         Polyarthra         5         6         7         6         6000         68.8           Image: Synchaetidae         Polyarthra         5         6         7         6         6000         68.8           Image: Synchaetidae         Daphina         Image: Synchaetidae         Image: Synchaetidae

Total Organisms per L	Rotifera	%	Cladocera	%	Copepoda	%
1227	1212	98.8%	15	1.2%	58	4.7%

## Site: Mt Komble Lake South Station

Data: 1/20/11

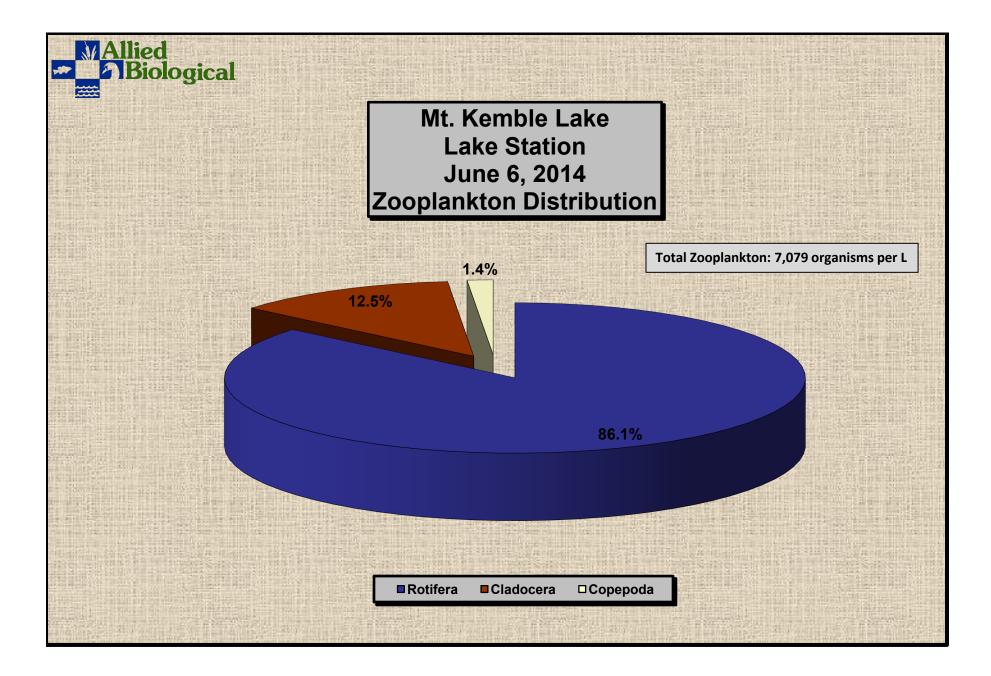


# Zooplankton Count Results



Mt. Kemb	le Lake		Date: 6/6/14							nogicai
					Replicate		Total/3	x1000 mL	Water	# organisms
Group	Order	Family	Genus	A	В	С	(# per mL)	( = 1 L)	sampled (L)	per L
Rotifera	Ploima	Asplanchnidae	Asplanchna	3	2	3	2.67	2,667	68.8	39
		Brachionidae	Brachionus	9	8	14	10.33	10,333	68.8	150
			Keratella	177	212	232	207.00	207,000	68.8	3009
			Kellicottia	80	67	49	65.33	65,333	68.8	950
		Synchaetidae	Polyarthra	48	35	38	121.00	121,000	68.8	1759
			Synchaeta	4	5	10	6.33	6,333	68.8	92
		Trichocercidae	Trichocerca			1	0.33	333	68.8	5
	Flosculariacea	Conochilidae	Conochilus	2	7	2	3.67	3,667	68.8	53
		Testudinellidae	Filinia	4	4		2.67	2,667	68.8	39
									Total:	6095
Cladocera	Cladocera									
		Daphniidae	Daphnia	66	51	65	60.67	60667	68.8	882
									Total:	882
Copepoda										
	Cyclopoida	Cyclopoidae	Cyclopoid nauplii	7	8	6	7.00	7000	68.8	102
									Total:	102

Total Organisms per L	Rotifera	%	Cladocera	%	Copepoda	%
7079	6095	86.1%	882	12.5%	102	1.4%

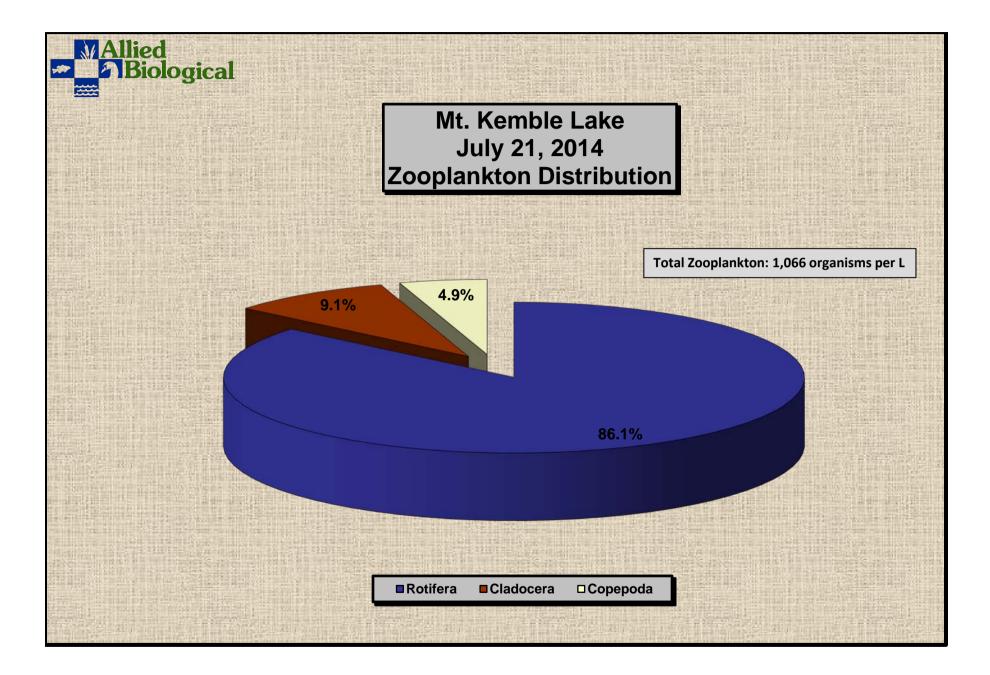


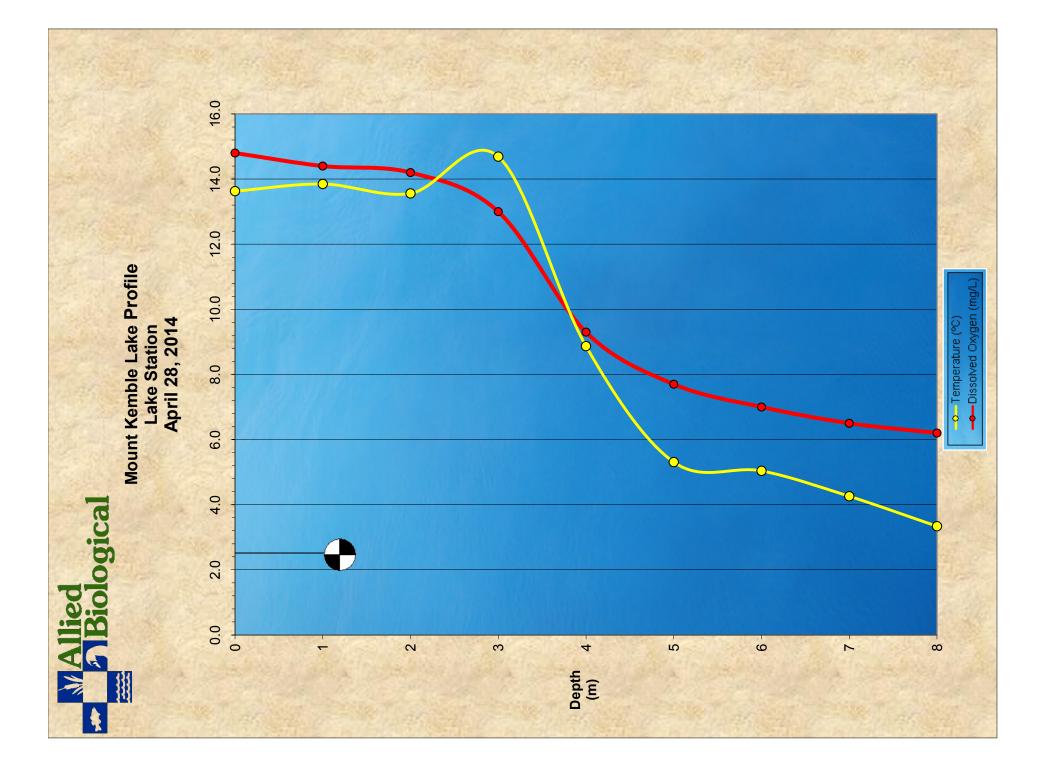
# Zooplankton Count Results

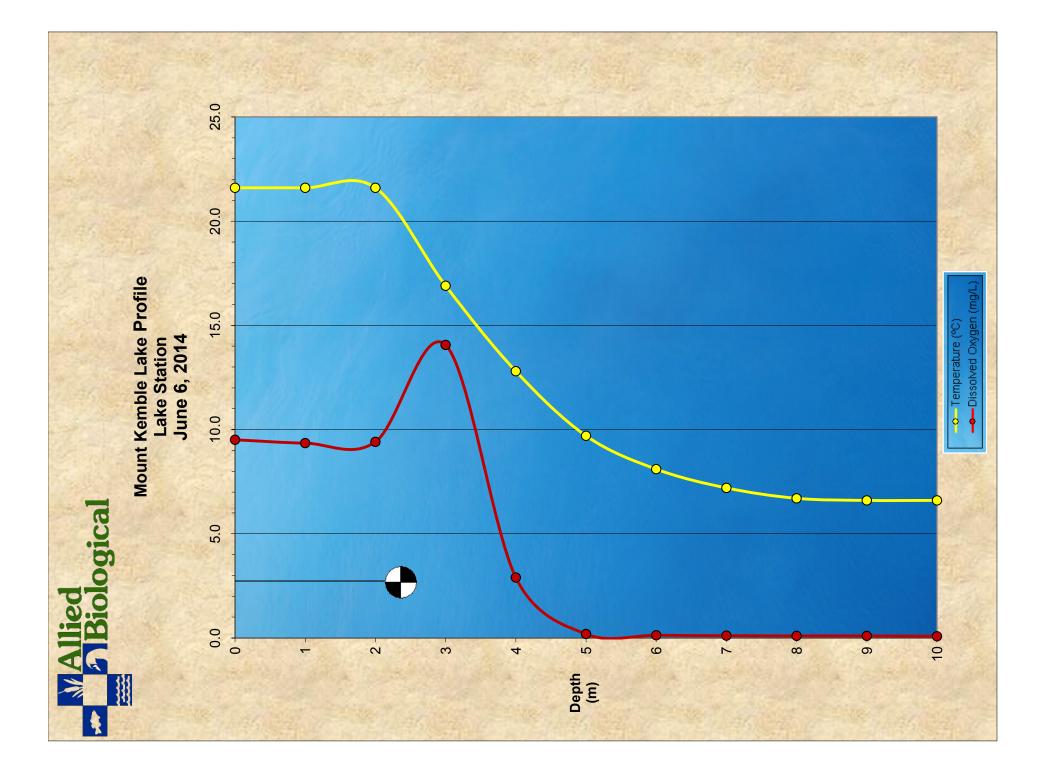


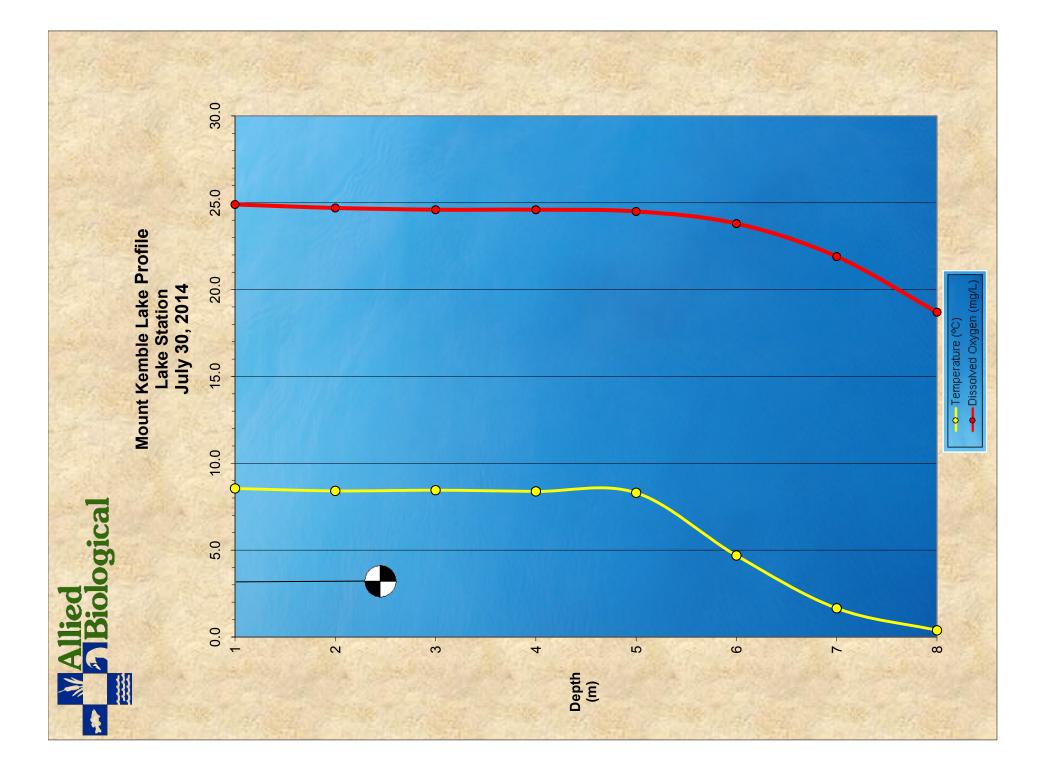
Mt. Kemb	e Lake		Date: 7/30/14							nogicui
					Replicate		Total/3	x1000 mL	Water	# organisms
Group	Order	Family	Genus	Α	В	С	(# per mL)	(=1L)	sampled (L)	per L
Rotifera	Ploima	Asplanchnidae	Asplanchna				0.00	0	68.8	0
		Brachionidae	Brachionus	3	7	6	5.33	5,333	68.8	78
			Keratella	21	28	25	24.67	24,667	68.8	359
			Kellicottia			1	0.33	333	68.8	5
		Synchaetidae	Polyarthra	16	17	14	47.00	47,000	68.8	683
			Synchaeta	5	1	6	4.00	4,000	68.8	58
		Gastropidae	Ascomorpha				0.00	0	68.8	0
		Trichocercidae	Trichocerca	1	1	1	1.00	1,000	68.8	15
	Flosculariacea	Conochilidae	Conochilus				0.00	0	68.8	0
									Total:	1197
Cladocera	Cladocera									
		Daphniidae	Ceriodaphnia				0.00	0	68.8	0
		Bosminidae	Bosmina	8	11	7	8.67	8667	68.8	126
									Total:	126
Copepoda										
	Cyclopoida	Cyclopoidae	Cyclopoid nauplii	2	6	6	4.67	4667	68.8	68
									Total:	68

Total Organisms per L	Rotifera	%	Cladocera	%	Copepoda	%
1391	1197	86.1%	126	9.1%	68	4.9%









APL

AQUA PRO-TECH LABORATORIES Certified Environmental Testing

# Analytical Results Summary MT. KEMBLE LAKE

Client: APL Order ID:	Allied Biological 4040816			ntact: eived:	Chris 4/28/1			
Sample ID/Analysis	Method	Prepared	Analyzed	Result	Qual	MDL	RL	Units
4040816-01 (Lake)	NOF	RTH STATION		Collected:				
General Chemistry								
Phosphorus	4500PE	5/7/14 11:00	5/7/14 11:00	0.0300			0.0100	mg/L
Nitrate	EPA 300	4/29/14 12:44	4/29/14 12:48	0.900			0.200	mg/L
Specific conductance	SM 2510B	4/30/14 11:00	4/30/14 11:00	320			1.00	umhos/cm
Total Suspended Solids	SM 2540D	5/1/14 11:47	5/1/14 11:47	ND			3.00	mg/L
Ammonia as N	SM 4500 NH3 D	5/5/14 15:00	5/5/14 15:00	ND			0.200	mg/L
4040816-02 (Lake)	LAK	E STATION SURF	ACE	Collected:	4/28/2	2014	10:25	
General Chemistry								
Phosphorus	4500PE	5/7/14 11:00	5/7/14 11:00	0.0300			0.0100	mg/L
Nitrate	EPA 300	4/29/14 12:44	4/29/14 12:48	0.900			0.200	mg/L
Specific conductance	SM 2510B	4/30/14 11:00	4/30/14 11:00	310			1.00	umhos/cm
Total Suspended Solids	SM 2540D	5/1/14 11:47	5/1/14 11:47	ND			3.00	mg/L
Ammonia as N	SM 4500 NH3 D	5/5/14 15:00	5/5/14 15:00	ND			0.200	mg/L
4040816-03 (Lake)	LAK	E STATION BOTT	ОМ	Collected:	4/28/2	2014	10:30	
General Chemistry								
Phosphorus	4500PE	5/7/14 11:00	5/7/14 11:00	0.0300			0.0100	mg/L
Nitrate	EPA 300	4/29/14 12:44	4/29/14 12:48	0.950			0.200	mg/L
Specific conductance	SM 2510B	4/30/14 11:00	4/30/14 11:00	317			1.00	umhos/cm
Total Suspended Solids	SM 2540D	5/1/14 11:47	5/1/14 11:47	ND			3.00	mg/L
Ammonia as N	SM 4500 NH3 D	5/5/14 15:00	5/5/14 15:00	ND			0.200	mg/L
4040816-04 (Lake)	OU	TLET STATION		Collected:	4/28/2	2014	11:25	
General Chemistry								
Phosphorus	4500PE	5/7/14 11:00	5/7/14 11:00	0.0300			0.0100	mg/L
4040816-05 (Lake)	INL	ET STATION		Collected:	4/28/2	2014	11:15	
		1		1				
General Chemistry				the second s				

# **Analytical Results Summary**

Mt. Kenble Lake

Client: APL Order ID:	Allied Biological 4060193			ntact: ceived:		5 Doyle 4 12:45		
Sample ID/Analysis	Method	Prepared	Analyzed	Result	Qual	MDL	RL	Units
4060193-01 (Lake)	Sou	th Station Surface		Collected	: 6/6/2	014 9:10	)	
General Chemistry								
Phosphorus	4500PE	6/10/14 14:00	6/10/14 14:00	0.0300			0.0100	mg/L
Nitrate	EPA 300	6/6/14 17:14	6/6/14 17:17	0.980			0.200	mg/L
Specific conductance	SM 2510B	6/9/14 15:00	6/9/14 15:00	290			1.00	umhos/cm
Total Suspended Solids	SM 2540D	6/11/14 11:11	6/11/14 11:11	ND			3.00	mg/L
Ammonia as N	SM 4500 NH3 D	6/12/14 15:00	6/12/14 15:00	ND			0.200	mg/L
4060193-02 (Lake)	Nor	th Station		Collected	6/6/2	2014 9:20	)	
General Chemistry			6/10/14 14:00		: 6/6/2	2014 9:20		
General Chemistry Phosphorus	4500PE EPA 300	6/10/14 14:00	6/10/14 14:00 6/6/14 17:17	0.0500	: 6/6/2	2014 9:20	0.0100	mg/L
General Chemistry Phosphorus Nitrate	4500PE		6/6/14 17:17	0.0500 0.980	: 6/6/2	014 9:20	0.0100 0.200	mg/L
General Chemistry Phosphorus Nitrate Specific conductance	4500PE EPA 300	6/10/14 14:00 6/6/14 17:14		0.0500	: 6/6/2	2014 9:20	0.0100 0.200 1.00	mg/L umhos/cm
General Chemistry Phosphorus Nitrate Specific conductance Total Suspended Solids	4500PE EPA 300 SM 2510B	6/10/14 14:00 6/6/14 17:14 6/9/14 15:00	6/6/14 17:17 6/9/14 15:00	0.0500 0.980 300	: 6/6/2	2014 9:20	0.0100 0.200	mg/L
	4500PE EPA 300 SM 2510B SM 2540D SM 4500 NH3 D	6/10/14 14:00 6/6/14 17:14 6/9/14 15:00 6/11/14 11:11	6/6/14 17:17 6/9/14 15:00 6/11/14 11:11	0.0500 0.980 300 ND		2014 9:20	0.0100 0.200 1.00 3.00 0.200	mg/L umhos/cm mg/L
General Chemistry Phosphorus Nitrate Specific conductance Total Suspended Solids Ammonia as N	4500PE EPA 300 SM 2510B SM 2540D SM 4500 NH3 D	6/10/14 14:00 6/6/14 17:14 6/9/14 15:00 6/11/14 11:11 6/12/14 15:00	6/6/14 17:17 6/9/14 15:00 6/11/14 11:11	0.0500 0.980 300 ND ND			0.0100 0.200 1.00 3.00 0.200	mg/L umhos/cm mg/L
General Chemistry Phosphorus Nitrate Specific conductance Total Suspended Solids Ammonia as N 4060193-03 (Lake)	4500PE EPA 300 SM 2510B SM 2540D SM 4500 NH3 D	6/10/14 14:00 6/6/14 17:14 6/9/14 15:00 6/11/14 11:11 6/12/14 15:00	6/6/14 17:17 6/9/14 15:00 6/11/14 11:11	0.0500 0.980 300 ND ND			0.0100 0.200 1.00 3.00 0.200	mg/L umhos/cm mg/L mg/L
General Chemistry Phosphorus Nitrate Specific conductance Total Suspended Solids Ammonia as N 4060193-03 (Lake) General Chemistry	4500PE EPA 300 SM 2510B SM 2540D SM 4500 NH3 D Sou	6/10/14 14:00 6/6/14 17:14 6/9/14 15:00 6/11/14 11:11 6/12/14 15:00 th Station Bottom	6/6/14 17:17 6/9/14 15:00 6/11/14 11:11 6/12/14 15:00	0.0500 0.980 300 ND ND Collected			0.0100 0.200 1.00 3.00 0.200	mg/L umhos/cm mg/L mg/L mg/L
General Chemistry Phosphorus Nitrate Specific conductance Total Suspended Solids Ammonia as N 4060193-03 (Lake) General Chemistry Phosphorus Nitrate	4500PE EPA 300 SM 2510B SM 2540D SM 4500 NH3 D Sou	6/10/14 14:00 6/6/14 17:14 6/9/14 15:00 6/11/14 11:11 6/12/14 15:00 th Station Bottom 6/10/14 14:00	6/6/14 17:17 6/9/14 15:00 6/11/14 11:11 6/12/14 15:00 6/10/14 14:00	0.0500 0.980 300 ND ND Collected			0.0100 0.200 1.00 3.00 0.200	mg/L umhos/cm mg/L mg/L mg/L
General Chemistry Phosphorus Nitrate Specific conductance Total Suspended Solids Ammonia as N 4060193-03 (Lake) General Chemistry Phosphorus	4500PE EPA 300 SM 2510B SM 2540D SM 4500 NH3 D Sou 4500PE EPA 300	6/10/14 14:00 6/6/14 17:14 6/9/14 15:00 6/11/14 11:11 6/12/14 15:00 th Station Bottom 6/10/14 14:00 6/6/14 17:14	6/6/14 17:17 6/9/14 15:00 6/11/14 11:11 6/12/14 15:00 6/10/14 15:00 6/10/14 14:00 6/6/14 17:17	0.0500 0.980 300 ND ND Collected 0.0800 ND			0.0100 0.200 1.00 3.00 0.200 5 5 0.0100 0.200	mg/L umhos/cm mg/L mg/L

ND - Indicates compound analyzed for but not detected

J - Indicates estimated value

B - Indicates compound found in associated blank

E - Concentration exceeds highest calibration standard

D - Indicates result is based on a dilution P - Greater than 25% diff. between 2 GC columns. MDL - Minimum detection limit RL - Reporting limit

**RO-TECH LABORATORIES** d Environmental Testing

## **Analytical Results Summary** MT KEMBLE

Client: APL Order ID:	Allied Biological 4070942			ntact: ceived:		Doyle 4 12:50		
Sample ID/Analysis	Method	Prepared	Analyzed	Result	Qual	MDL	RL	Units
4070942-01 (Lake)	NO	RTH STATION		Collected	7/30/2	2014 12:	:11	
General Chemistry		12						
Phosphorus	4500PE	8/6/14 10:00	8/6/14 10:00	0.0100			0.0100	mg/L
Nitrate	EPA 300	7/30/14 16:05	7/30/14 16:05	ND			0.200	mg/L
Specific conductance	SM 2510B	8/7/14 11:00	8/7/14 11:00	291			1.00	umhos/cm
Total Suspended Solids	SM 2540D	8/4/14 18:16	8/4/14 18:16	ND			3.00	mg/L
Ammonia as N	SM 4500 NH3 D	8/5/14 14:00	8/5/14 14:00	ND			0.200	mg/L
4070942-02 (Lake)	SO	JTH STATION SU	RFACE	Collected	7/30/2	2014 11:	:00	
General Chemistry		25 2	~		7/30/2	2014 11:		
General Chemistry Phosphorus	4500PE	8/6/14 10:00	8/6/14 10:00	0.0200	7/30/2	2014 11:	0.0100	mg/L
General Chemistry Phosphorus Nitrate	4500PE EPA 300	8/6/14 10:00 7/30/14 16:05	8/6/14 10:00 7/30/14 16:05	0.0200 ND	7/30/2	2014 11:	<b>0.0100</b> 0.200	mg/L
General Chemistry Phosphorus Nitrate Specific conductance	4500PE	8/6/14 10:00 7/30/14 16:05 8/7/14 11:00	8/6/14 10:00 7/30/14 16:05 8/7/14 11:00	0.0200 ND 291	7/30/2	2014 11:	0.0100 0.200 1.00	mg/L umhos/cm
General Chemistry Phosphorus Nitrate	4500PE EPA 300 SM 2510B	8/6/14 10:00 7/30/14 16:05	8/6/14 10:00 7/30/14 16:05	0.0200 ND	7/30/2	2014 11:	<b>0.0100</b> 0.200	
General Chemistry Phosphorus Nitrate Specific conductance Total Suspended Solids	4500PE EPA 300 SM 2510B SM 2540D SM 4500 NH3 D	8/6/14 10:00 7/30/14 16:05 8/7/14 11:00 8/4/14 18:16	8/6/14 10:00 7/30/14 16:05 8/7/14 11:00 8/4/14 18:16 8/5/14 14:00	0.0200 ND 291 ND		2014 11:	0.0100 0.200 1.00 3.00 0.200	mg/L umhos/cm mg/L
General Chemistry Phosphorus Nitrate Specific conductance Total Suspended Solids Ammonia as N	4500PE EPA 300 SM 2510B SM 2540D SM 4500 NH3 D	8/6/14 10:00 7/30/14 16:05 8/7/14 11:00 8/4/14 18:16 8/5/14 14:00	8/6/14 10:00 7/30/14 16:05 8/7/14 11:00 8/4/14 18:16 8/5/14 14:00	0.0200 ND 291 ND ND			0.0100 0.200 1.00 3.00 0.200	mg/L umhos/cm mg/L
General Chemistry Phosphorus Nitrate Specific conductance Total Suspended Solids Ammonia as N 4070942-03 (Lake)	4500PE EPA 300 SM 2510B SM 2540D SM 4500 NH3 D	8/6/14 10:00 7/30/14 16:05 8/7/14 11:00 8/4/14 18:16 8/5/14 14:00	8/6/14 10:00 7/30/14 16:05 8/7/14 11:00 8/4/14 18:16 8/5/14 14:00	0.0200 ND 291 ND ND			0.0100 0.200 1.00 3.00 0.200	mg/L umhos/cm mg/L mg/L
General Chemistry Phosphorus Nitrate Specific conductance Total Suspended Solids Ammonia as N 4070942-03 (Lake) General Chemistry Phosphorus	4500PE EPA 300 SM 2510B SM 2540D SM 4500 NH3 D	8/6/14 10:00 7/30/14 16:05 8/7/14 11:00 8/4/14 18:16 8/5/14 14:00 JTH STATION BO	8/6/14 10:00 7/30/14 16:05 8/7/14 11:00 8/4/14 18:16 8/5/14 14:00	0.0200 ND 291 ND ND Collected:			0.0100 0.200 1.00 3.00 0.200	mg/L umhos/cm mg/L mg/L
General Chemistry Phosphorus Nitrate Specific conductance Total Suspended Solids Ammonia as N 4070942-03 (Lake) General Chemistry	4500PE EPA 300 SM 2510B SM 2540D SM 4500 NH3 D SOU	8/6/14 10:00 7/30/14 16:05 8/7/14 11:00 8/4/14 18:16 8/5/14 14:00 JTH STATION BO 8/6/14 10:00	8/6/14 10:00 7/30/14 16:05 8/7/14 11:00 8/4/14 18:16 8/5/14 14:00 TTOM	0.0200 ND 291 ND ND Collected: 0.0200			0.0100 0.200 1.00 3.00 0.200 :10 :10	mg/L umhos/cm mg/L mg/L mg/L
General Chemistry Phosphorus Nitrate Specific conductance Total Suspended Solids Ammonia as N 4070942-03 (Lake) General Chemistry Phosphorus Nitrate	4500PE EPA 300 SM 2510B SM 2540D SM 4500 NH3 D SOU	8/6/14 10:00 7/30/14 16:05 8/7/14 11:00 8/4/14 18:16 8/5/14 14:00 JTH STATION BO 8/6/14 10:00 7/30/14 16:05	8/6/14 10:00 7/30/14 16:05 8/7/14 11:00 8/4/14 18:16 8/5/14 14:00 TTOM 8/6/14 10:00 7/30/14 16:05	0.0200 ND 291 ND ND Collected: 0.0200 ND			0.0100 0.200 1.00 3.00 0.200 10	mg/L umhos/cm mg/L mg/L mg/L

ND - Indicates compound analyzed for but not detected J - Indicates estimated value B - Indicates compound found in associated blank E - Concentration exceeds highest calibration standard

D - Indicates result is based on a dilution P - Greater than 25% diff, between 2 GC columns. MDL - Minimum detection limit RL - Reporting limit